

Nuove proprietà Chimico-fisiche dell'acqua



V. Elia, E. Napoli & M. Niccoli



Pulsatilla

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Nux Vomica



Sponsors :

• **REGIONE CAMPANIA**

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Arnica Montana



32. V.Elia, E.Napoli, M.Niccoli

“Calorimetric and Conductometric Titrations of Nanostructures of Water Molecules in Iteratively Filtered Water”

JTAC, 2011, Submitted

31. V.Elia, N. Marchettini, E.Napoli, E.Tiezzi

“Nanostructures of Water Molecules in Iteratively Filtered Water”

JOAM, 2011 Submitted

30. V.Elia, E.Napoli

“Nanostructures of Water Molecules in Iteratively Filtered Water”

IC-MAST, 2011(key Engineering materials) Accepted

29. L.Betti, V.Elia, E. Napoli ,G.Trebbi, M.Zurla, D.Nani ,M.Peruzzi, M.Brizzi

“Biological effects and physico-chemical properties of extremely diluted aqueous solutions as a function of aging-time”

FLS, 2011, In press

28. T.M.P.Cattaneo, S.Vero, E.Napoli, V.Elia

“Influence of Filtration Processes on Aqueous Nanostructures by NIR Spectroscopy”

JCHE, 2011 Online

27. V.Elia, L.Marrari, E.Napoli

“Aqueous Nanostrutture in water induced by Electromagnetic Field Emitted by EDS.

A Conductometric Study of Fullerene and Carbon Nanotube EDS.”

JTAC, 2011 Online

26.,V.Elia, E.Napoli, R.Rizzo

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25. M.Brizzi, V.Elia, G.Trebbi, D.Nani, M.Peruzzi, L.Betti

“The Efficacy of Ultramolecular Aqueous Dilutions on a Wheat Germination Model as a Function of Heat and Aging-Time”

eCAM, 2009, Doi:10.1093/ecam/nep217

24. T.M.P.Cattaneo, S.Vero, E.Napoli, V.Elia

“Studio degli effetti di processi fisici di filtrazione sulla formazione di nanostrutture acquose mediante spettroscopia NIR”

IV Simposio Italiano di Spettroscopia nel Vicino Infrarosso –Sestri Levante 2010

23. V.Elia, E.Napoli, M.Niccoli

“Thermodynamic Parameters for the Binding Process of the OH- Ion with the Dissipative Structures. Calorimetric and Conductometric Titrations”

Calorimetric and Conductometric Titrations

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22. V.Elia, E.Napoli

“Dissipative Structures in Extremely Diluted Solutions of Homeopathic Medicines. A Molecular Model based on Physico-Chemical and Gravimetric evidences”

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21. V.Elia, E.Napoli, M.Niccoli

“A Molecular Model of Interaction between of Extremely Diluted Solutions and NaOH Solutions Used as Titrant. Conductometric and pHmetric Titrations.”

Journal of Molecular Liquids, 2009, Vol.149, 45-50

20. C.M.Cacace, L.Elia, V.Elia, E.Napoli, M.Niccoli

“Conductometric and pHmetric Titrations of Extremely Diluted Solutions Using HCl Solutions as Titrant. A Molecular Model.”

Journal of Molecular Liquids, 2009, Vol.146, 122-126

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Dipartimento



19. E.Del Giudice, V.Elia, E. Napoli, A.Tedeschi
“The Role of Water in The Living Organisms”
Neural Biology, 2009, Vol.19, (4), 355-360

18. L.Ciavatta, V. Elia, E. Napoli, M. Niccoli
“ New Physico-Chemical Properties Of Extremely Diluted Solutions.
Electromotive Force Measurement of Galvanic Cells Sensible to the Activity of NaCl at 25°C
Journal of Solution chemistry , (2008) ,Vol. 37, 1037-1049

17. V.Elia,L.Elia,N. Marchettini, E.Napoli, M.Niccoli, E. Tiezzi
Physico-Chemical Properties of aqueous Extremely Diluted Solutions in relation to ageing
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16. V. Elia, E. Napoli, M.Niccoli
On the stability of extremely diluted aqueous solutions at the high ionic strength. A calorimetric study at 298K
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15. P.Belon, V. Elia, L. Elia, M. Montanino, E. Napoli, M. Niccoli
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14. V.Elia
“Response to Comments on “New Physico-Chemical Properties of Extremely Dilute Solutions. A Conductivity Study at 25°C in Relations to Ageing by Horatio R. Corti.
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13. V.Elia, E.Napoli, M.Niccoli, N. Marchettini, E. Tiezzi

New Physico-Chemical Properties of Extremely Dilute solutions. A Conductivity Study at 25°C in Relation to Ageing”.

Journal of Solution Chemistry, 2008, Vol.37, 85-96

12. V. Elia, E. Napoli,

“Strutture Dissipative nelle Soluzioni Estremamente Diluite della Medicina Omeopatica”

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11. V. Elia, E. Napoli, R. Germano

“The ‘memory of water’ an almost deciphered enigma. Dissipative structures in the extremely diluted aqueous solutions of the homeopathic medicine.

Homeopathy, (2007) 96, 163-169

10. V. Elia, L. Elia, E. Napoli, M. Niccoli

“Conductometric and calorimetric studies of serially diluted and agitated solutions: the dependance of intensive parameters on volume”

International Journal of Ecodynamics, (2006),Vol.1 (4), 361-372

9. V. Elia, L. Elia, M. Marchese, M. Montanino, E. Napoli, M. Niccoli, L. Nonatelli, F. Savarese

“Interaction of “extremely diluted solutions” with aqueous solutions of hydrochloric acid and sodium hydroxide. A calorimetric study.”

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8. V. Elia, L. Elia, M. Montanino, E. Napoli, M. Niccoli, L. Nonatelli

“Conductometric studies of the Serially Diluted and Agitated Solutions.On an anomalous effect that depends on the dilution process.”

Journal of Molecular Liquids , 2007, Vol. 135, 158-165

7. V. Elia, L. Elia, P. Cacace, E. Napoli, M. Niccoli, F. Savarese

“Extremely dilute solutions as multi-variable systems. A study of calorimetric and conductometric behaviour as function of the parameter time”

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6. V.Elia, M.Marchese, M.Montanino, E.Napoli, M.Niccoli, L.Nonatelli, A.Ramaglia
"Hydrohysteretic phenomena of "Extremely diluted solutions" induced by mechanical treatments. A calorimetric and conductometric study at 25 °C ".
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5. V.Elia, E.Napoli, M.Niccoli, L.Nonatelli, A. Ramaglia, E.Ventimiglia
"New Physico-Chemical Properties of Extremely Diluted Aqueous Solutions. A calorimetric and conductivity study at 25°C."
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4. V.Elia, M. Niccoli
"New Physico-Chemical Properties of Extremely Diluted Aqueous Solutions "
Journal of Thermal Analysis and Calorimetry, 2004, Vol. 75, 815-836

3. V.Elia, S.Baiano, I.Duro, E.Napoli, M.Niccoli, L.Nonatelli
Permanent Physico-chemical properties of extremely diluted aqueous solutions of Homeopathic Medicines.
Homeopathy, 2004, Vol 93, 144-150

2. V.Elia and M. Niccoli
"New physico-chemical properties of water induced by mechanical treatments. A Calorimetric study at 25°C"
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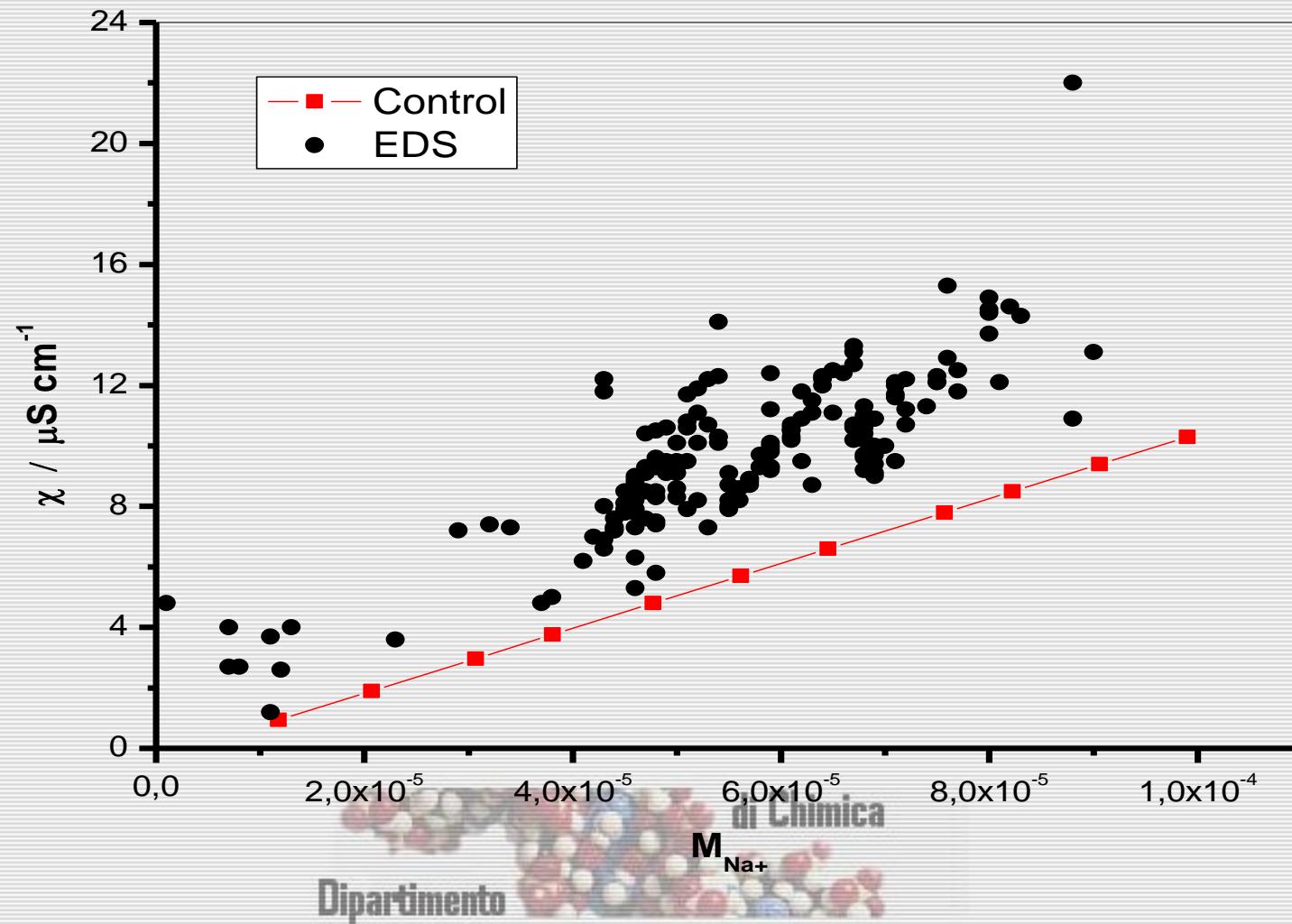
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"Thermodynamics of Extremely Diluted Aqueous Solutions"
Annals of the New York Academy of Sciences, (1999), Vol. 879, 241-248



“No one really understands water.
It’s embarrassing to admit it, but
the stuff that covers two-thirds of
our planet is still a mystery. Worse,
the more we look, the more the
problems accumulate: new
techniques probing deeper into the
molecular architecture of liquid
water are throwing up more
puzzles.” said P. Ball in 2008

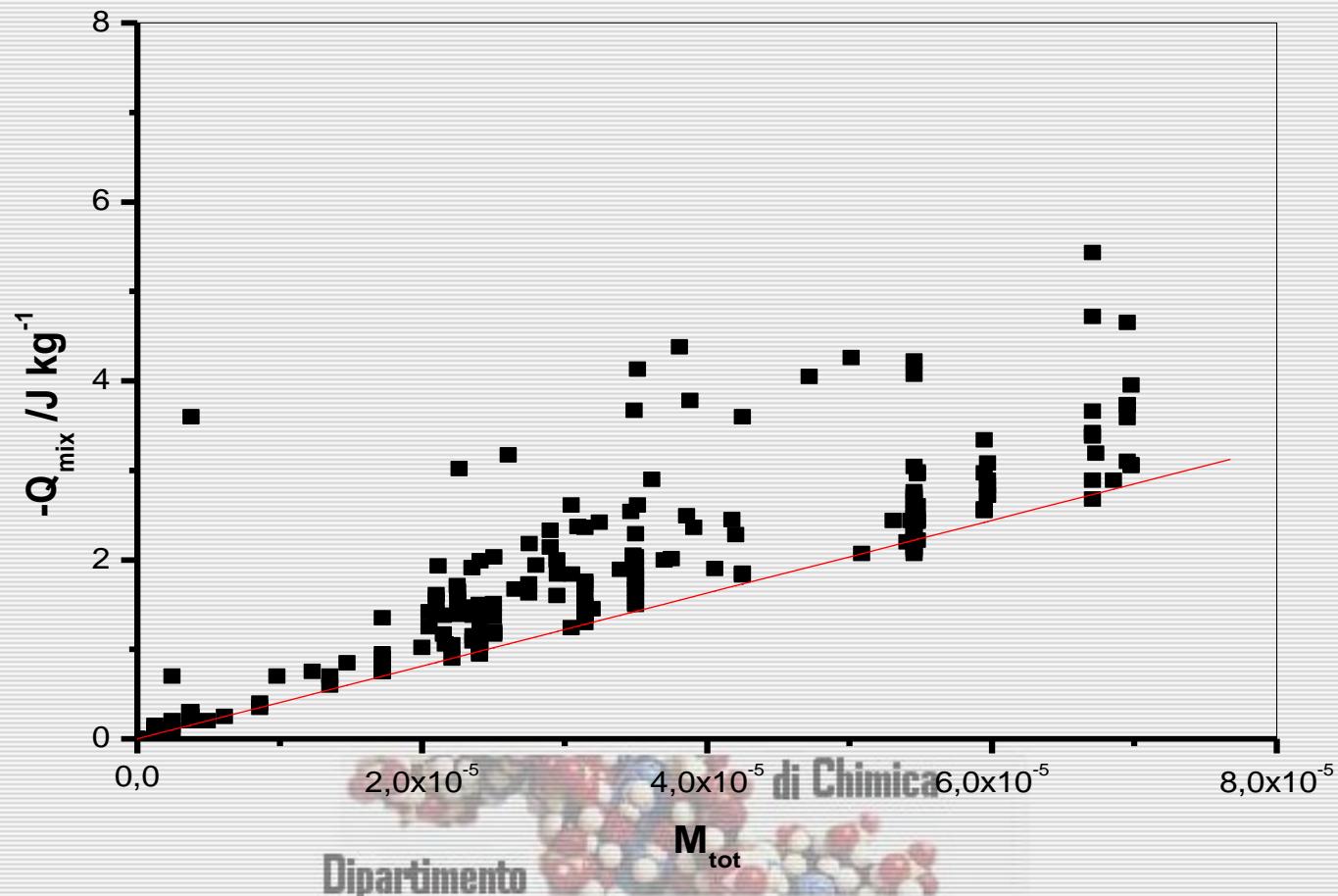
Ball P. Water: water-an enduring mystery. Nature 2008 Mar 20; 452(7185):291-2.

Specific Conductivity of EDS vs. Concentration of the Chemical Contents



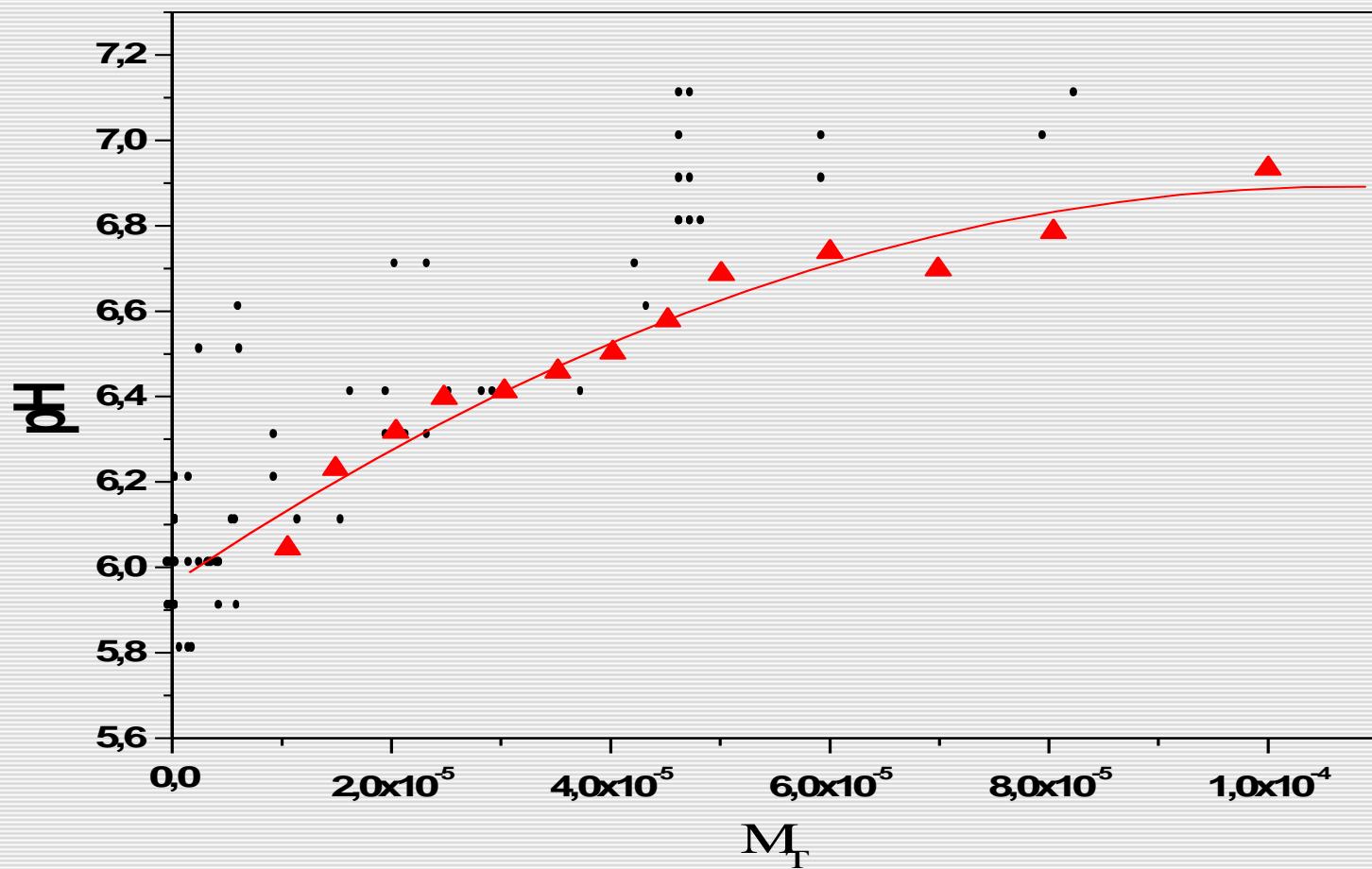


Heat of Mixing of EDS vs Concentration of the Chemical Contents





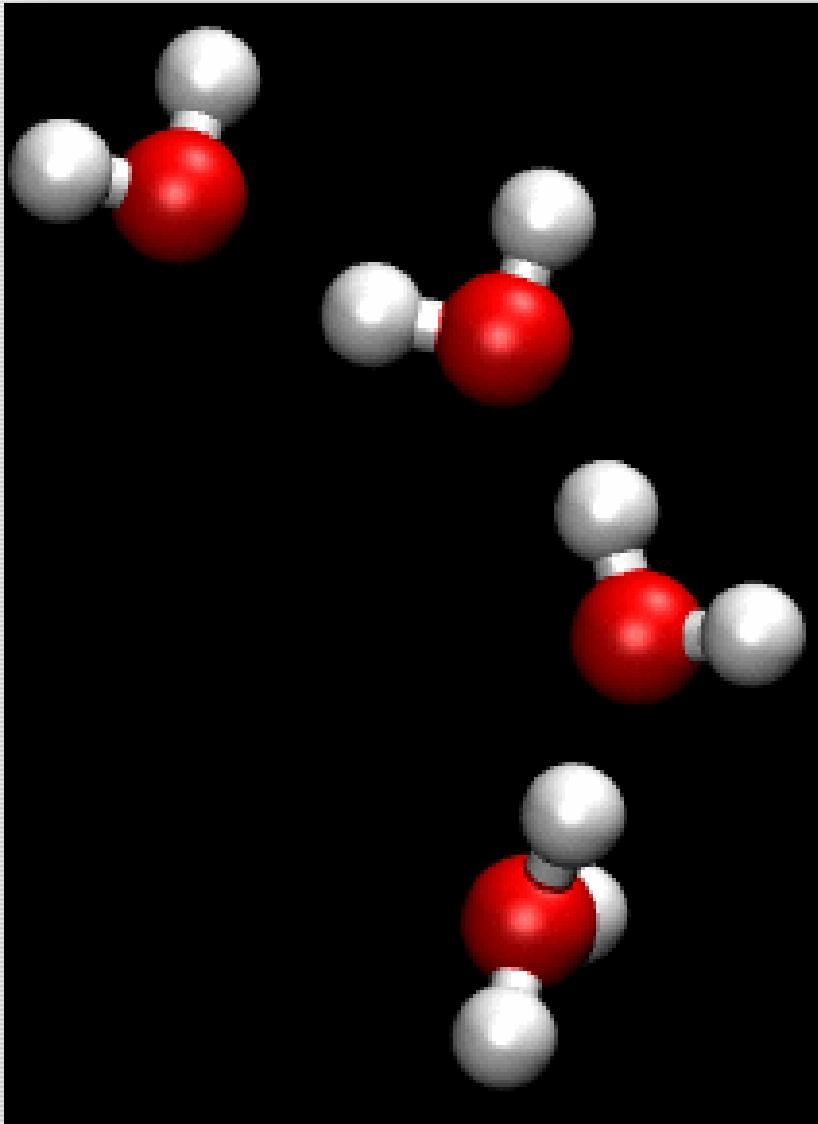
pH of EDS vs Concentration of the Chemical Contents

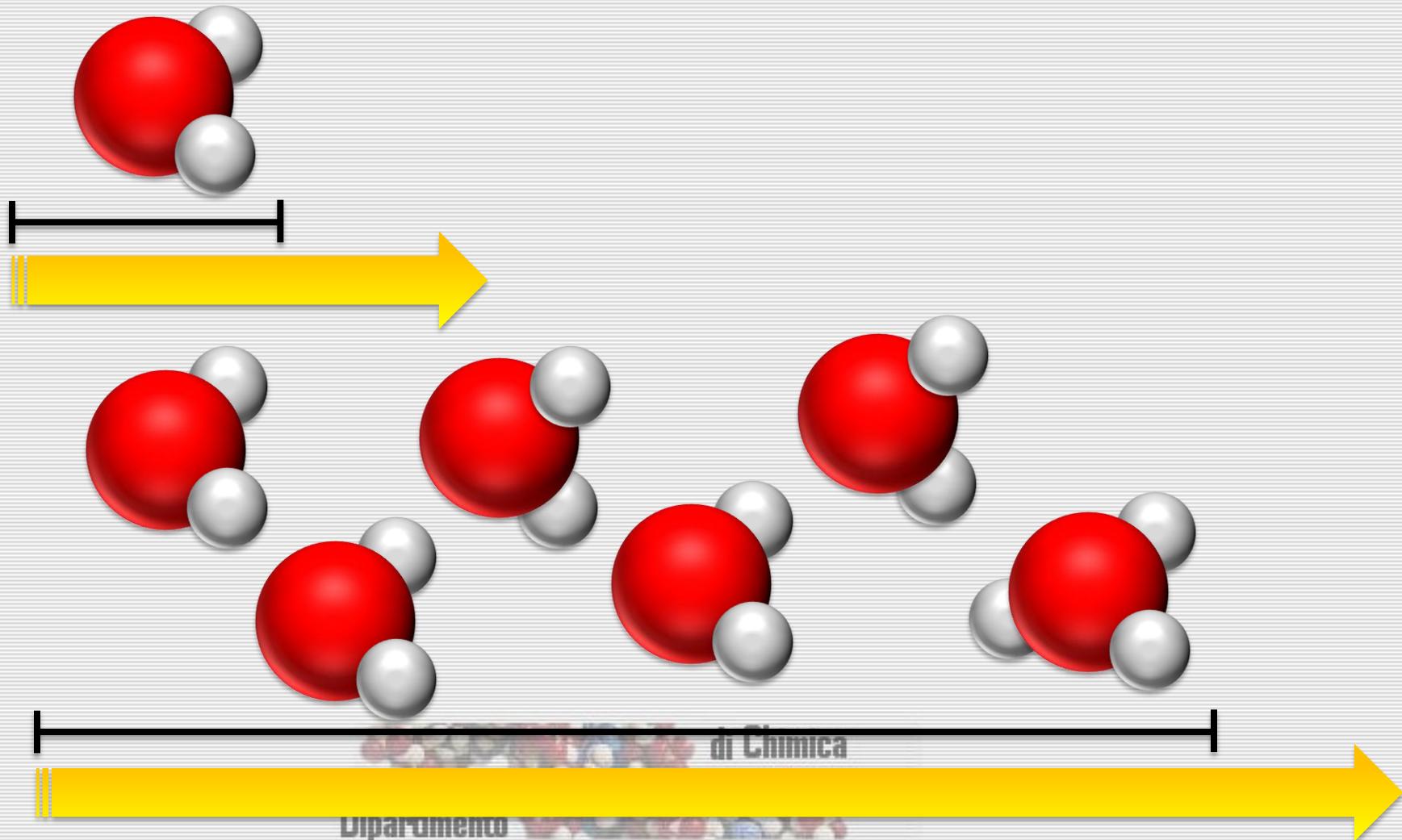


V. Elia, E. Napoli, R. Germano

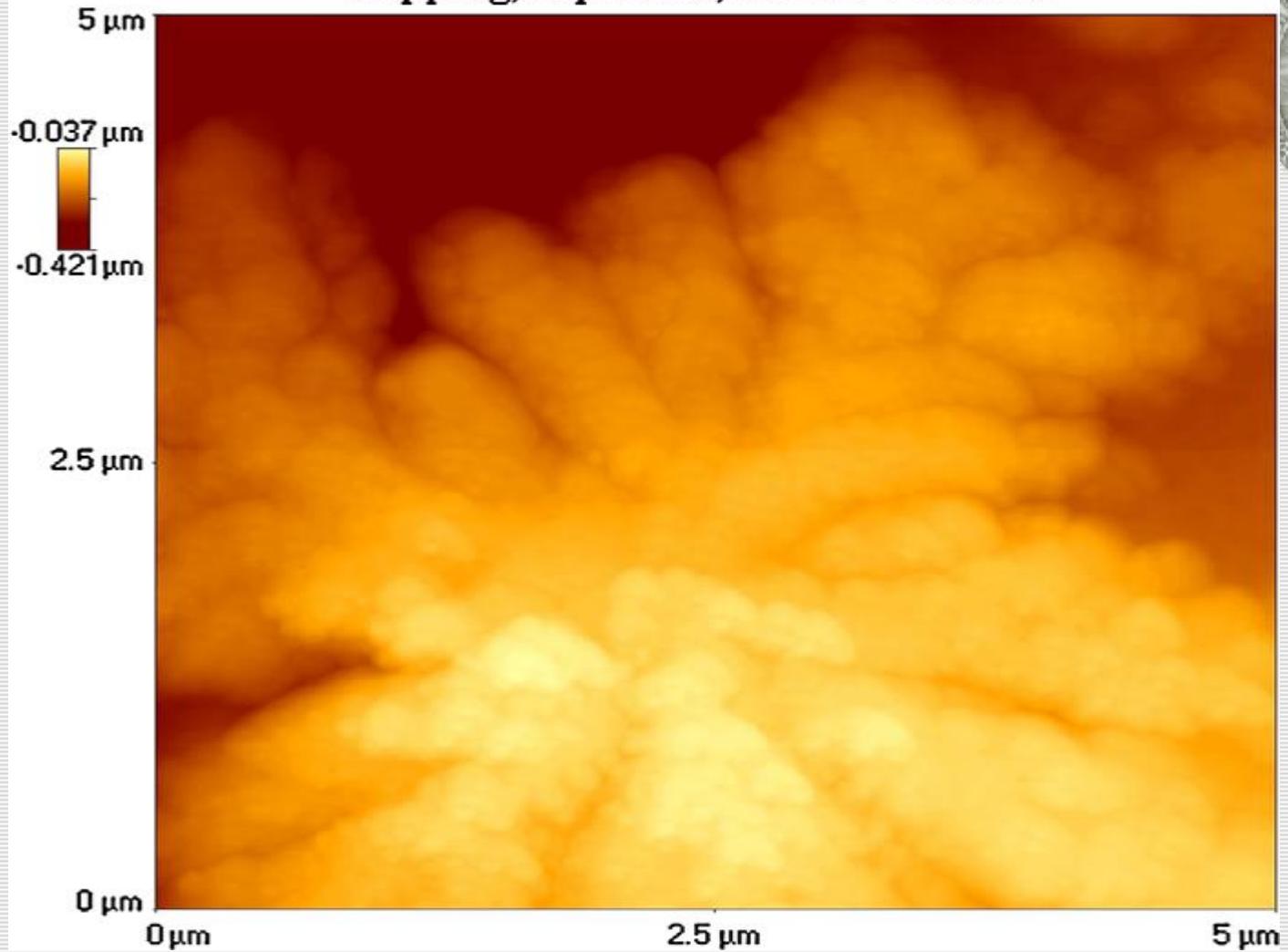
Homeopathy, (2007) 96, 163-169

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Tapping, Topo Frw, 1.7×10^{-7} diluent

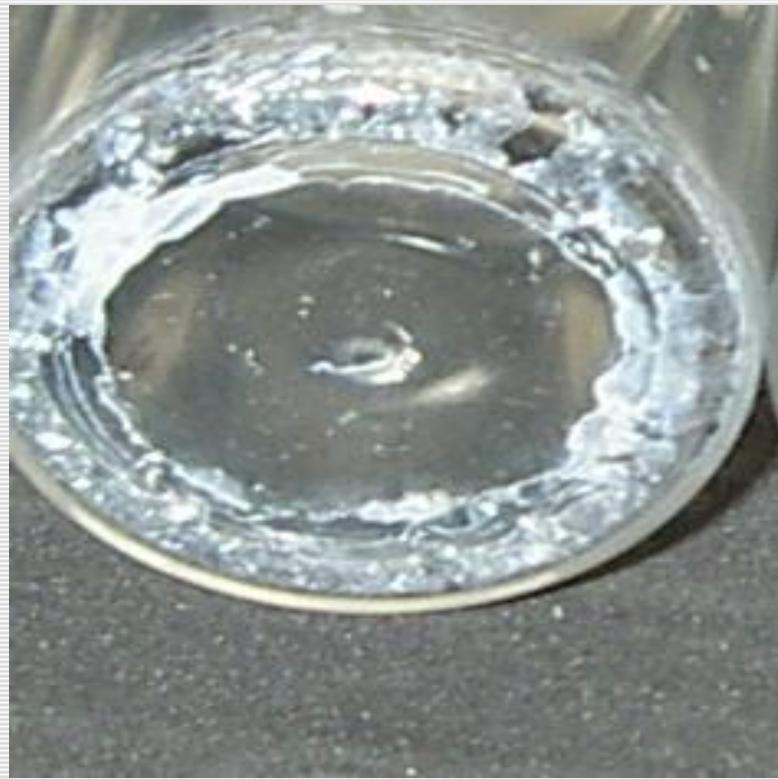


Physics Letters A – Shui Yin Lo, Xu Geng, David Gann

Evidence for the existence of stable-water-clusters at room temperature and normal pressure, 2009



Detail





	Na+	Xna	Xi	XE	mg Sp	mg Attesi	Δ
23	70,4	65,9	100	34,1	2,3	0,9	1,4
24	64,7	60,6	122	61,4	3,3	1,1	2,2
25	14,8	14,1	25	10,9	1,3	0,5	0,8
26	26,6	25,1	57	31,9	2,2	0,8	1,4
27	54,6	51,2	58	6,8	3,1	1,0	2,1
28	41,1	38,6	44	5,4	2,6	1,4	1,2
29	5,0	5,0	9,7	4,7	0,4	0,2	0,2
31	10,5	10,1	13	2,9	0,8	0,4	0,4
32	15,6	14,8	23	8,2	1,3	0,4	0,9
33	11,1	10,6	21	10,4	0,6	0,1	0,5
34	8,4	8,1	13	4,9	0,8	0,2	0,6
35	6,4	6,3	52	45,7	1,0	0,2	0,8
36	85,8	80,3	130	49,7	2,7	1,4	1,3

IJDN – V. Elia, E. Napoli

Dissipative Structures in Extremely Diluted Solution of Homeopathic Medicines. A Molecular Model based on Physico-Chemical and Gravimetric evidences, 2010, Vol 5, N°1, 39-48



Table 2

PV

M¹

Al ₂ O ₃	H ₂ BO ₃	H ₄ S _i O ₄	Na ₂ CO ₃	X _i ²	mg _{Expec}	mg _{Exper}	Δ _{mg}
0	1,8*10 ⁻³	1,3*10 ⁻⁴	5,5*10 ⁻⁴	68	3,1	3,1	0
0	8,2*10 ⁻⁵	0	4,2*10 ⁻⁵	8,5	0,2	0,3	0,1
0	6,1*10 ⁻⁴	4,3*10 ⁻⁵	1,8*10 ⁻⁴	18	1,0	1,2	0,2

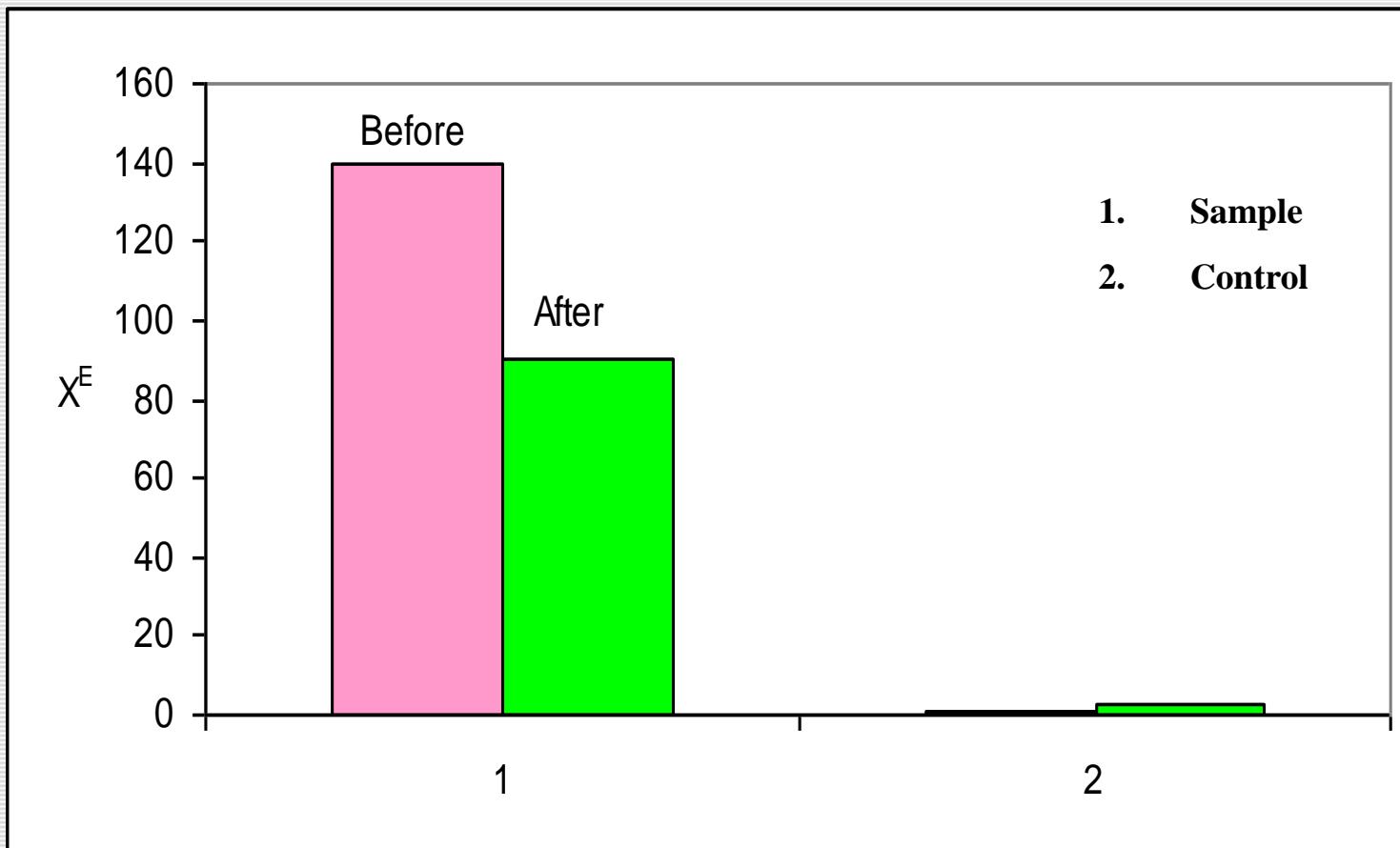
¹ Molarità in mol/L⁻¹ ²Initial Conductiviy in μS cm⁻¹,

IJDN – V. Elia, E. Napoli

Dissipative Structures in Extremely Diluted Solution of Homeopathic Medicines. A Molecular Model based on Physico-Chemical and Gravimetric evidences, 2010, Vol 5, N°1, 39-48



Evaporation



IJDN – V. Elia, E. Napoli

Dissipative Structures in Extremely Diluted Solution of Homeopathic Medicines. A Molecular Model based on Physico-Chemical and Gravimetric evidences, 2010, Vol 5, N°1, 39-48

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Dipartimento



Intervista a Luc Montagnier (1/2)



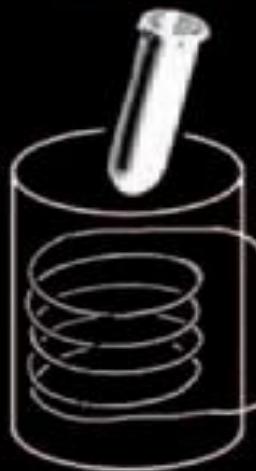


Intervista a Luc Montagnier (2/2)



Cattura dei segnali

Campione



X 500

**Software per l'analisi
dei segnali**



Solenoide sensore

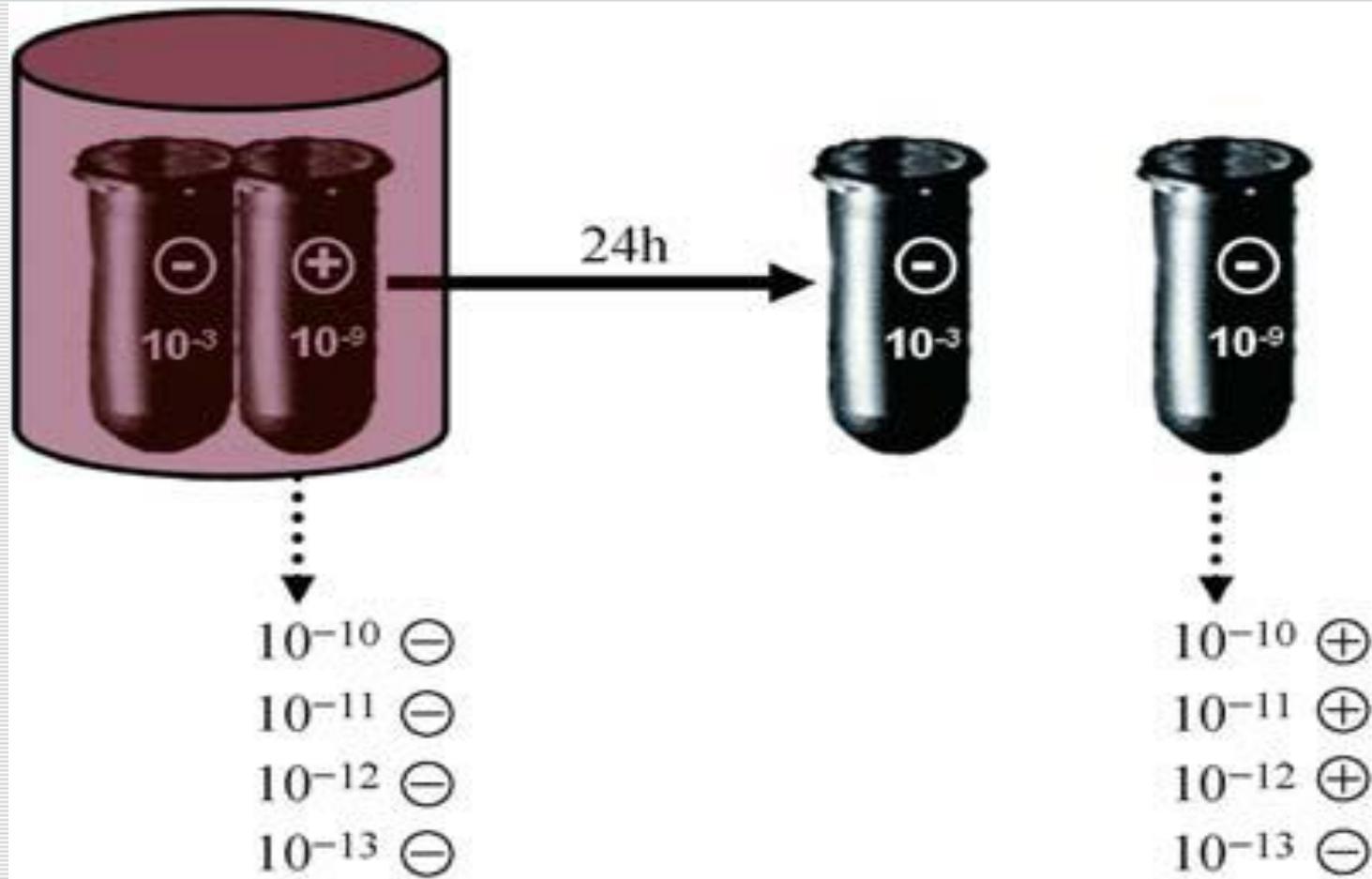
Amplificatore

Computer

Interdiscip Sci Comput Life Sci, L.Montagnier, J.Aissa, S.Ferris, J.L.Montagnier, C.Lavallée, 2009

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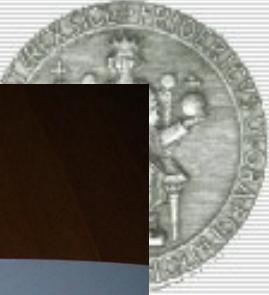
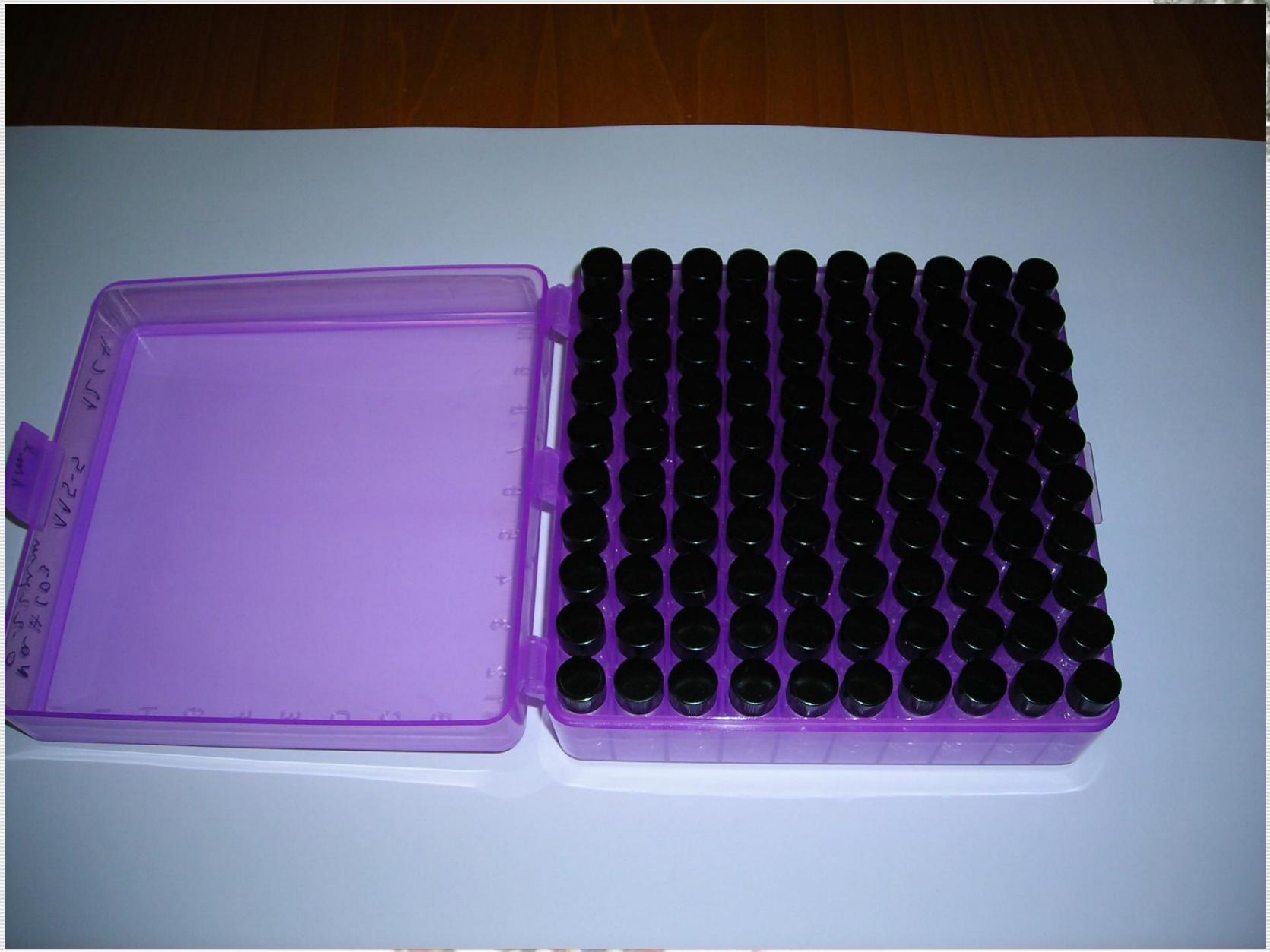
Dipartimento



Cross-talk between dilutions (from an *E.Coli* 0,1 μ filtrate) Montagnier 2009 Interdiscip Sci Comput Life Sci

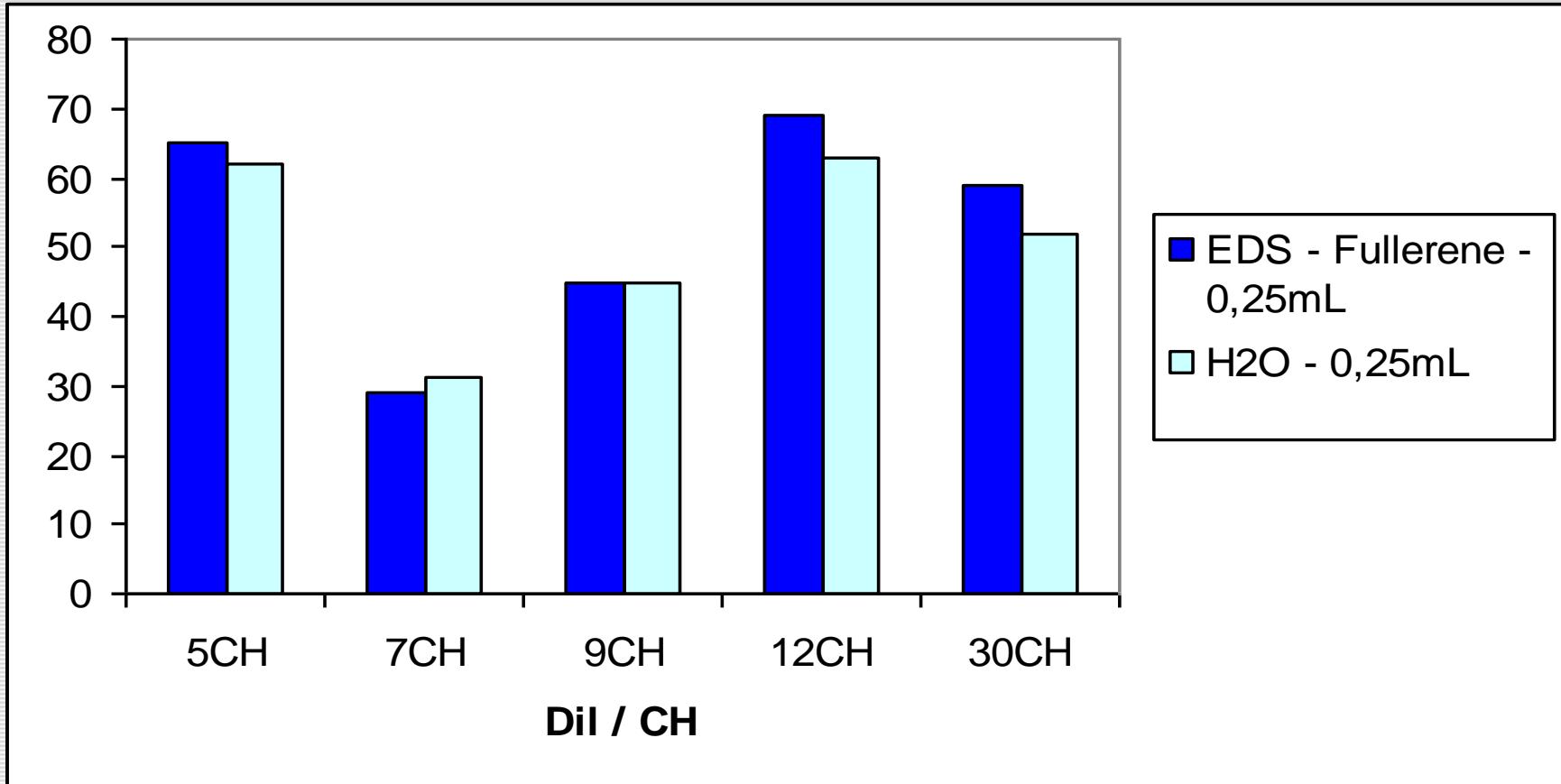


Electromagnetic Signals of Dissipative Structures and their ageing effects





Specific Electric Conductivity of EDS and H₂O at 135 days





Specific Electric Conductivity of EDS and H₂O at 541 days

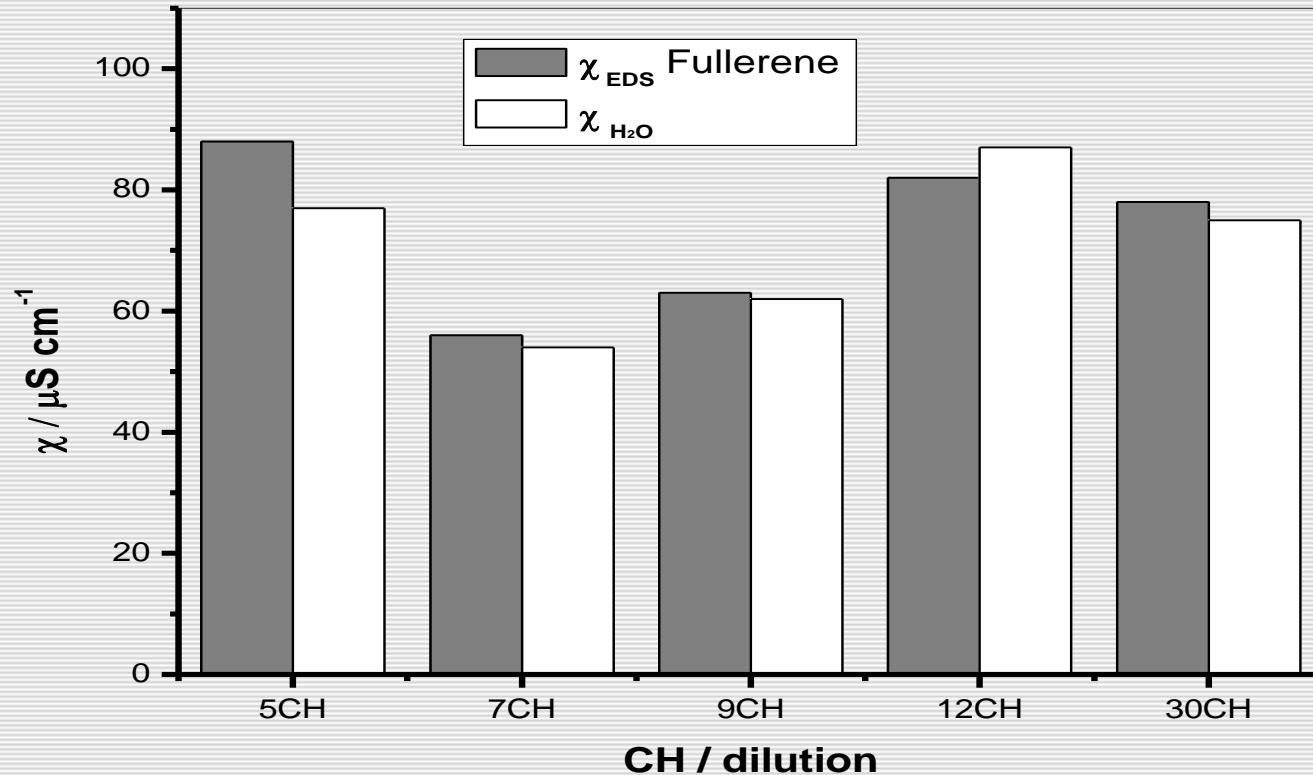


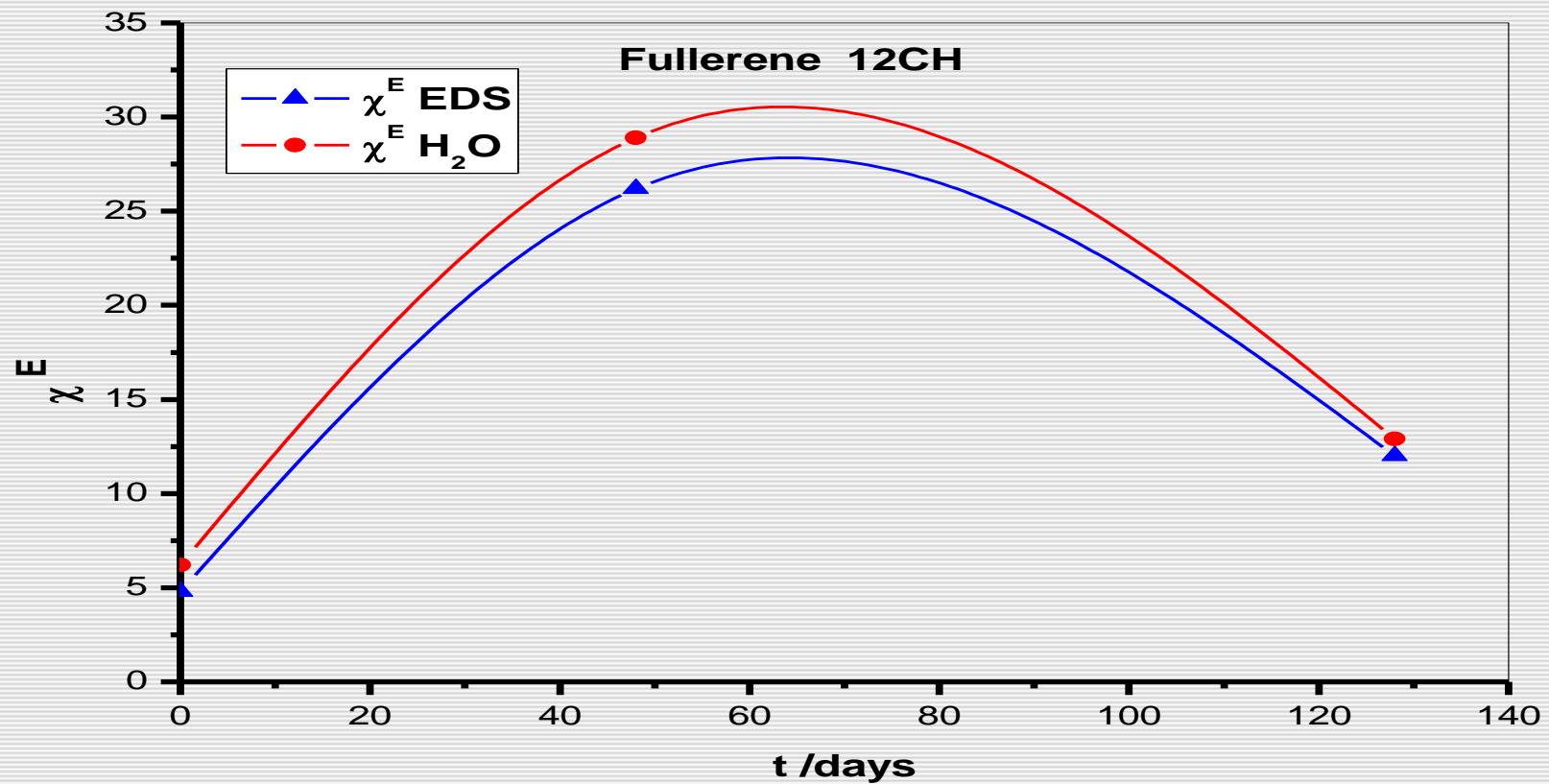
Fig.6 - Specific Electrical conductivity of EDS (Fullerene) and distilled H₂O at 541 days

V.Elia,L.Marrari, E.Napoli ,2011 JTAC , in press

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Excess Specific Electric Conductivity of EDS and H₂O vs t



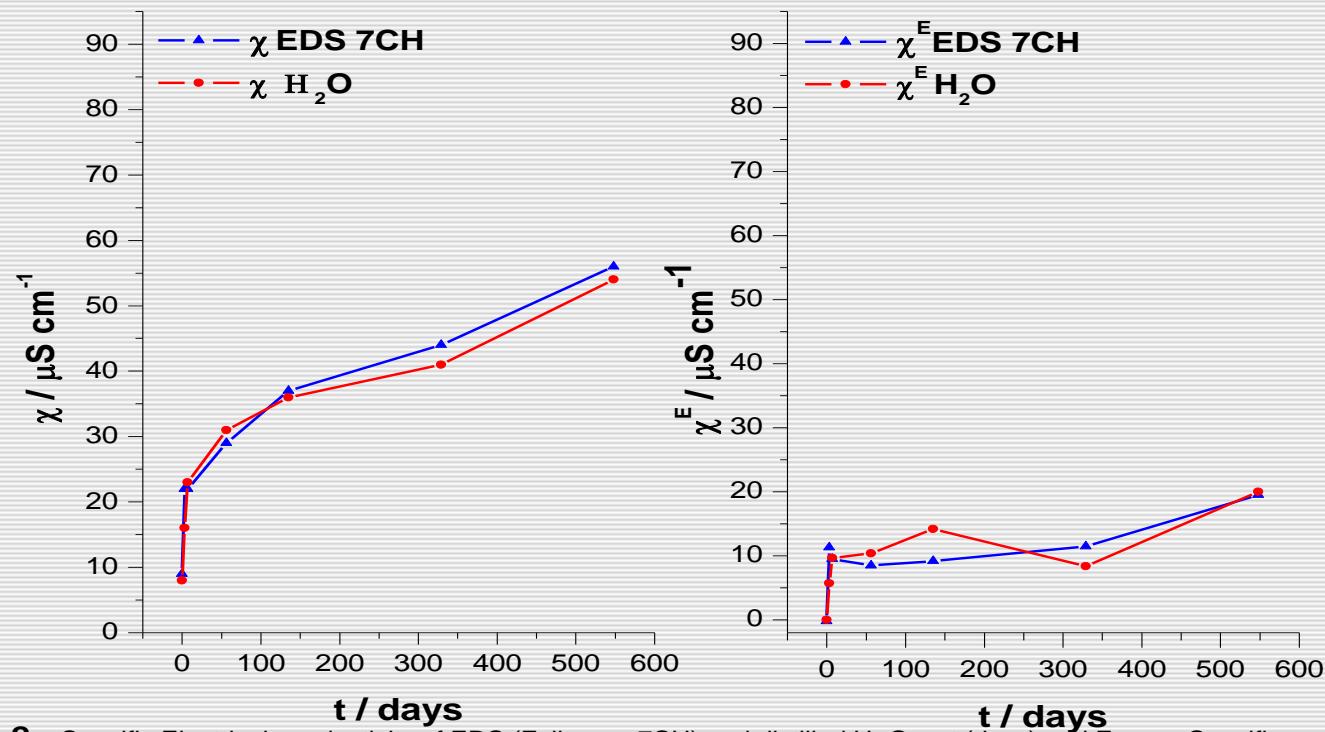


Fig.2 - Specific Electrical conductivity of EDS (Fullerene 7CH) and distilled H_2O vs t (days) and Excess Specific Electrical conductivity of EDS vs t (days)

JTAC - V. Elia,, L. Marrari, E. Napoli-
Aqueous Nanostructures In Water Induced By Electromagnetic Fields Emitted by EDS.
A Conductometric Study Of Fullerene And Carbon Nanotube EDS.2011, On line

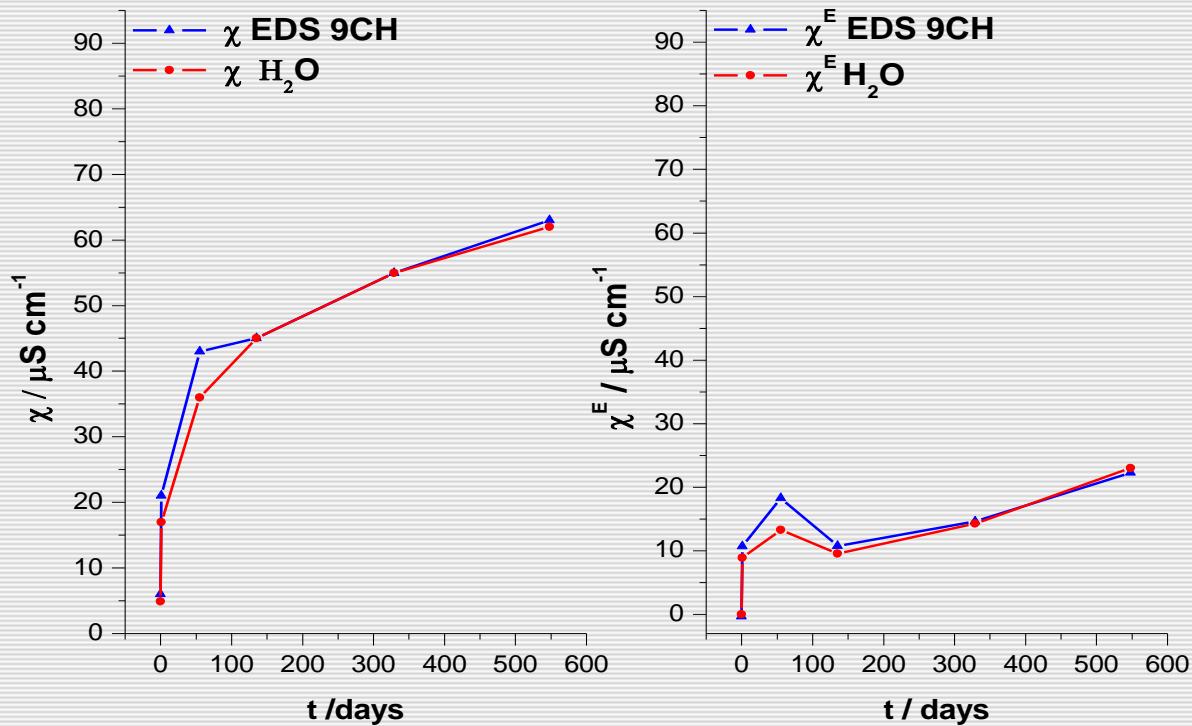
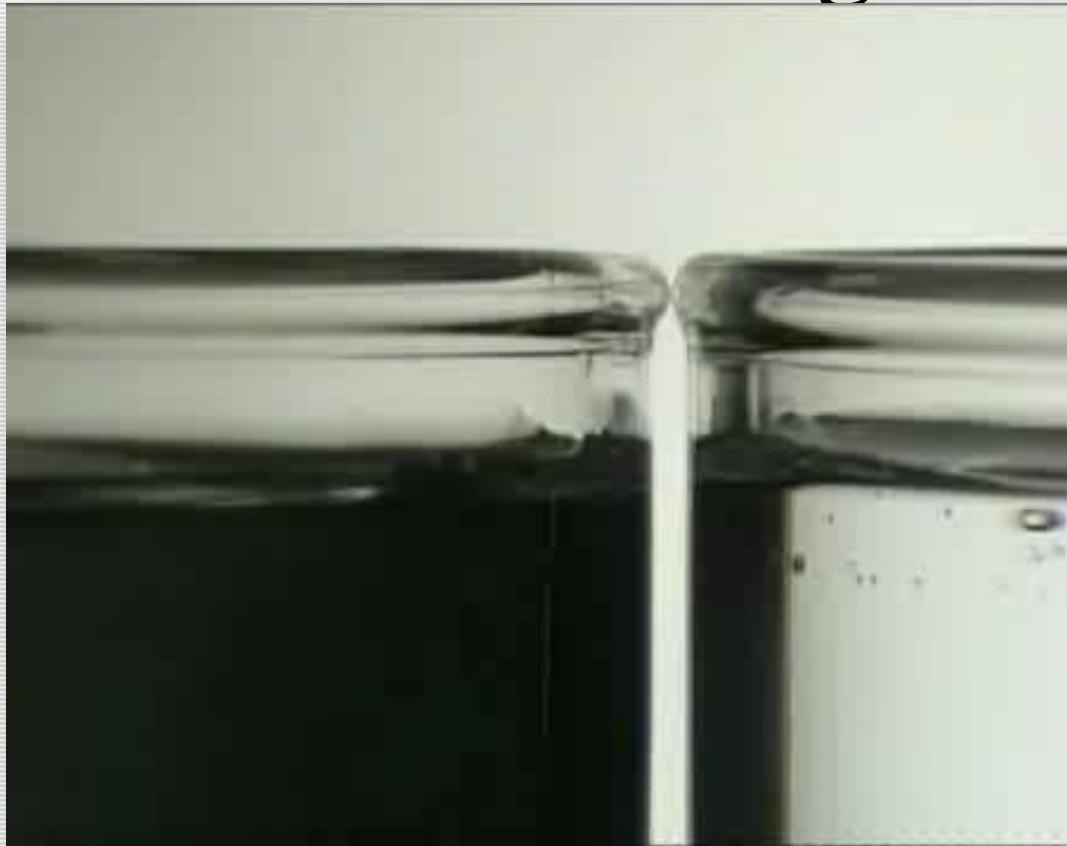


Fig.3 - Specific Electrical conductivity of EDS (Fullerene 9CH) and distilled H_2O vs t (days) and Excess Specific Electrical conductivity of EDS vs t (days)

*JTAC - V. Elia,, L. Marrari, E. Napoli-
Aqueous Nanostructures In Water Induced By Electromagnetic Fields Emitted by EDS.
A Conductometric Study Of Fullerene And Carbon Nanotube EDS.2011, In press*

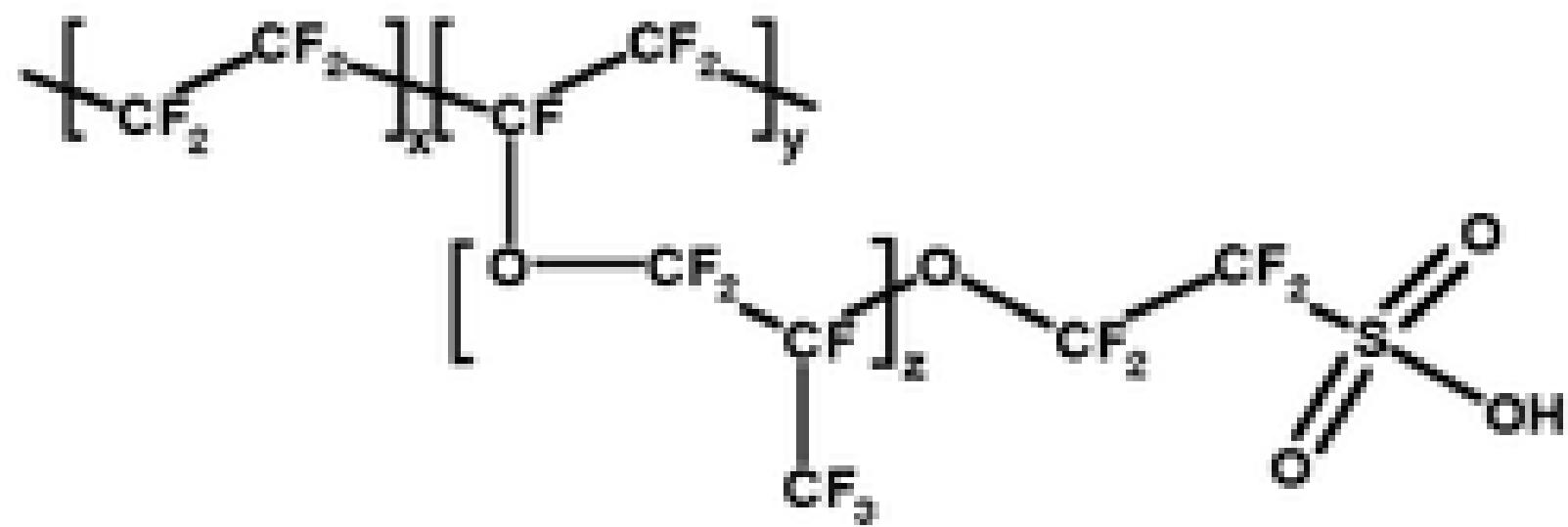


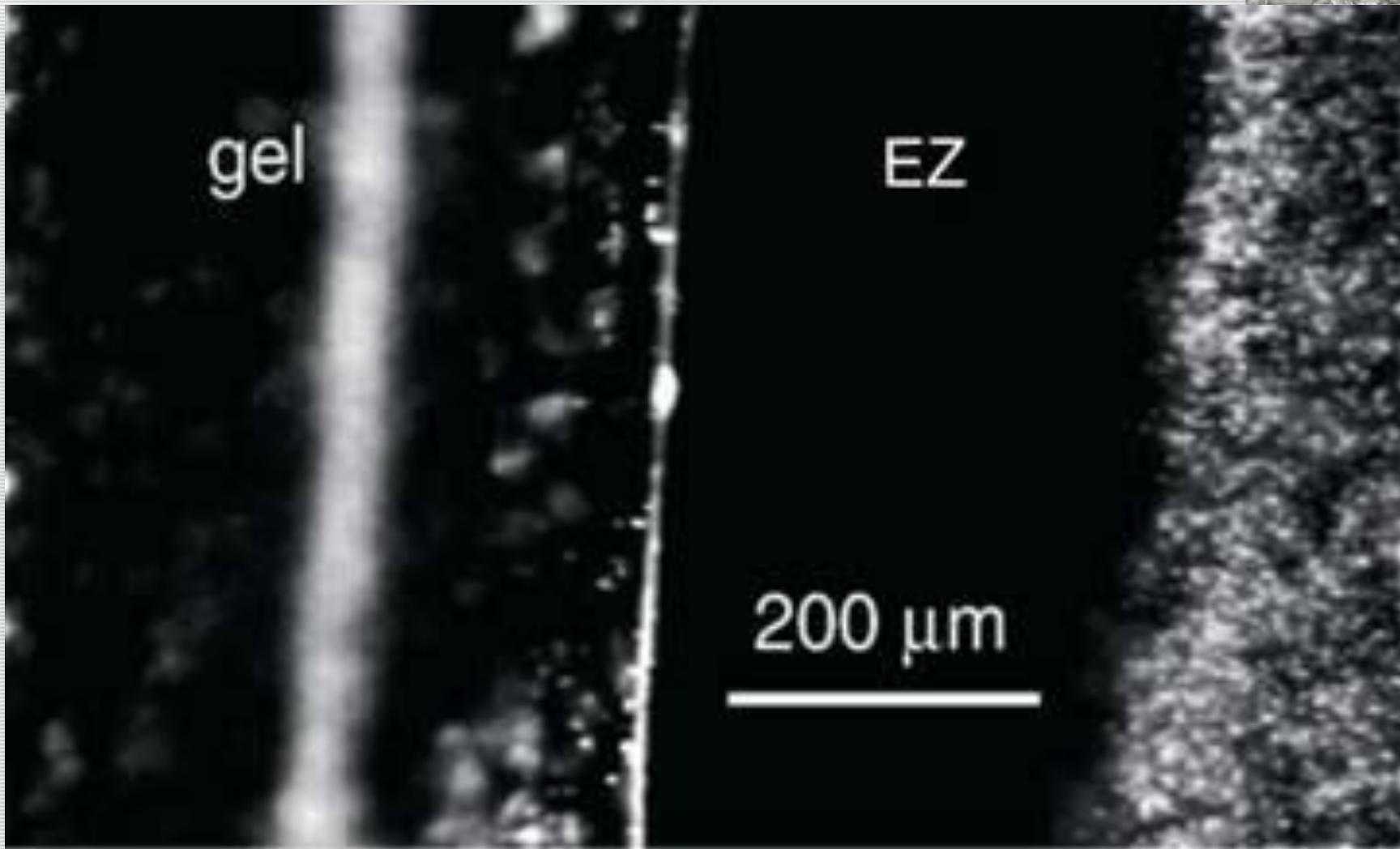
Water Bridge





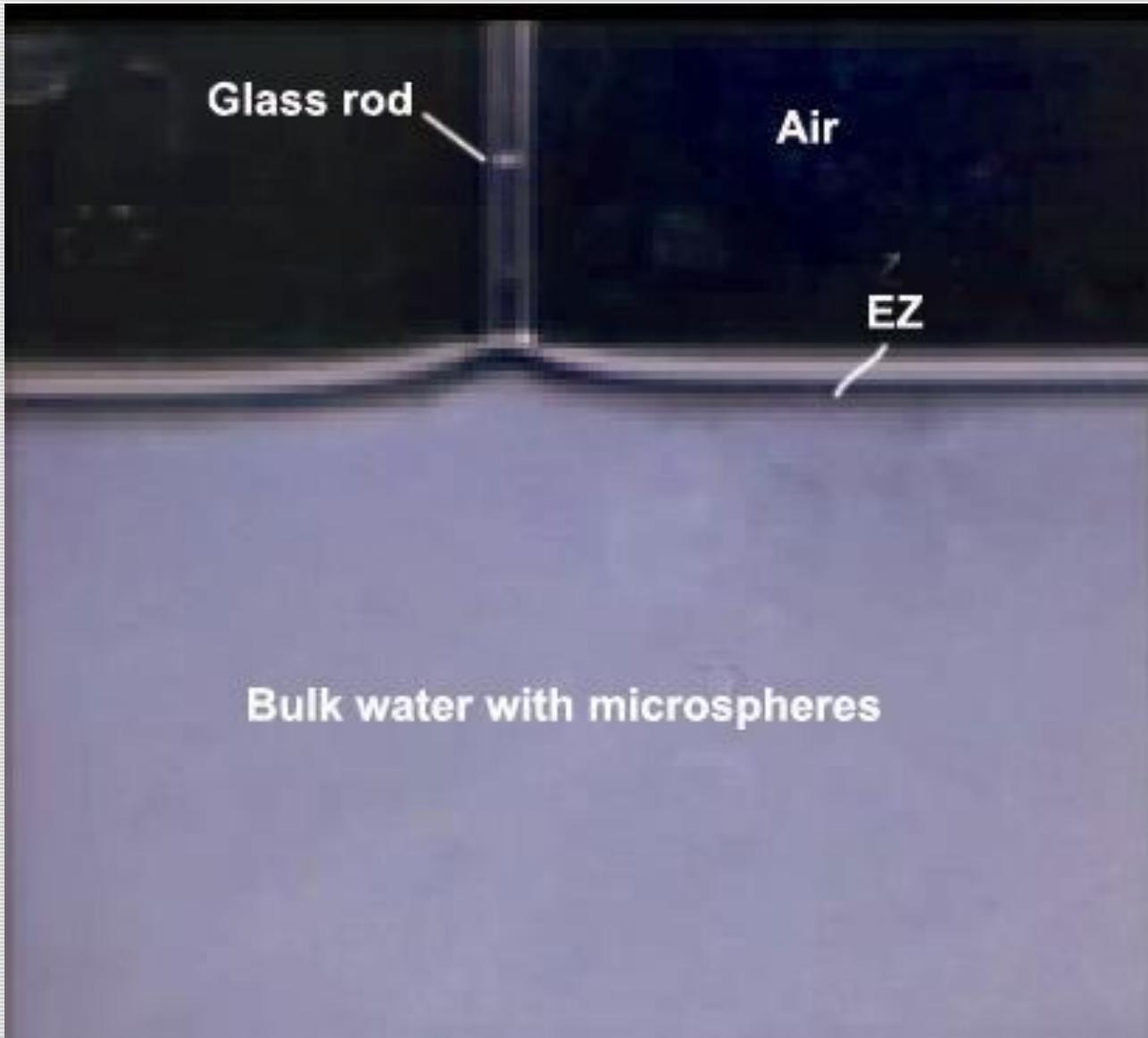
The Effect of Nafion on the Sovramolecular Structure of Water





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pH Sensitive Dye(s)

H⁺

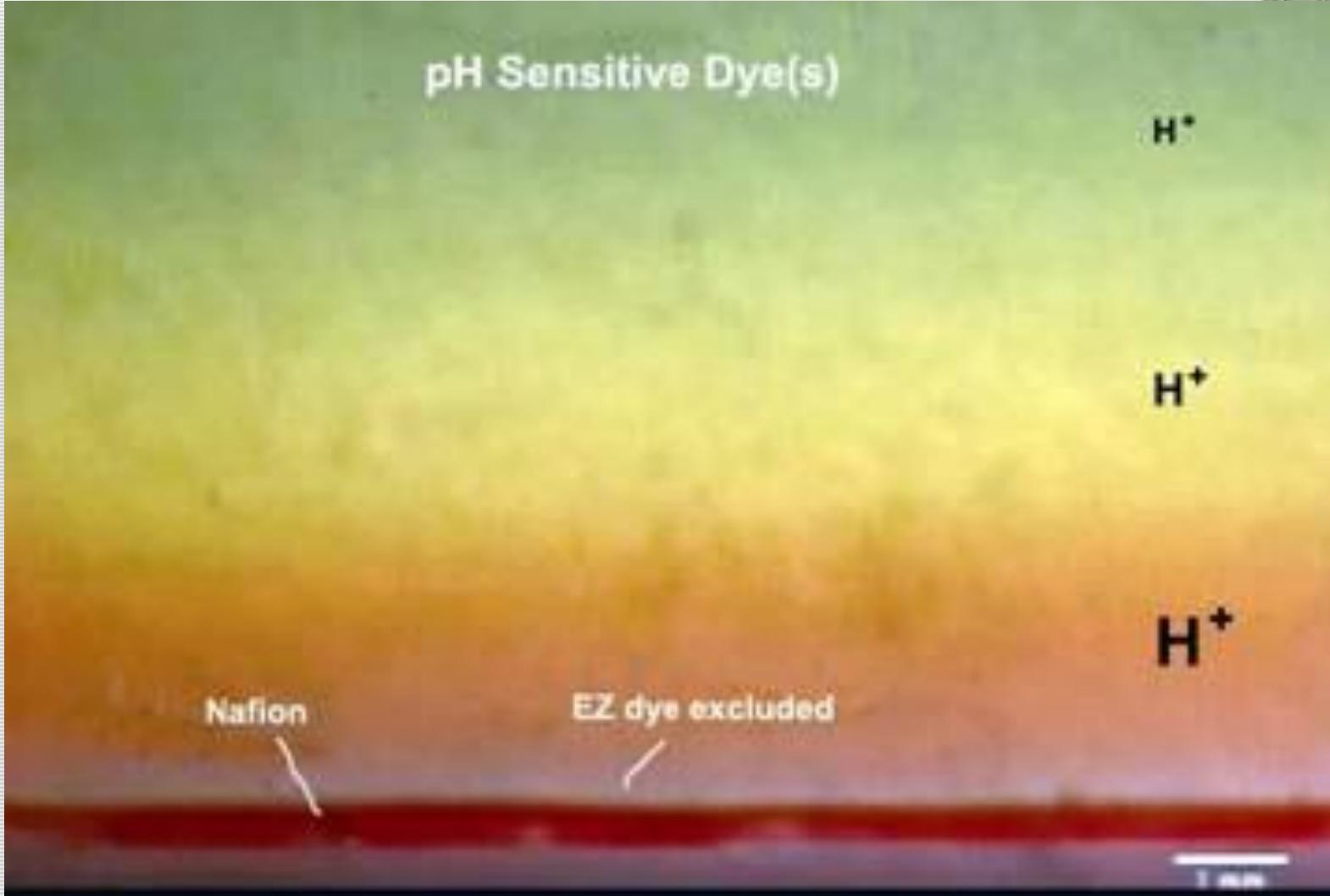
H⁺

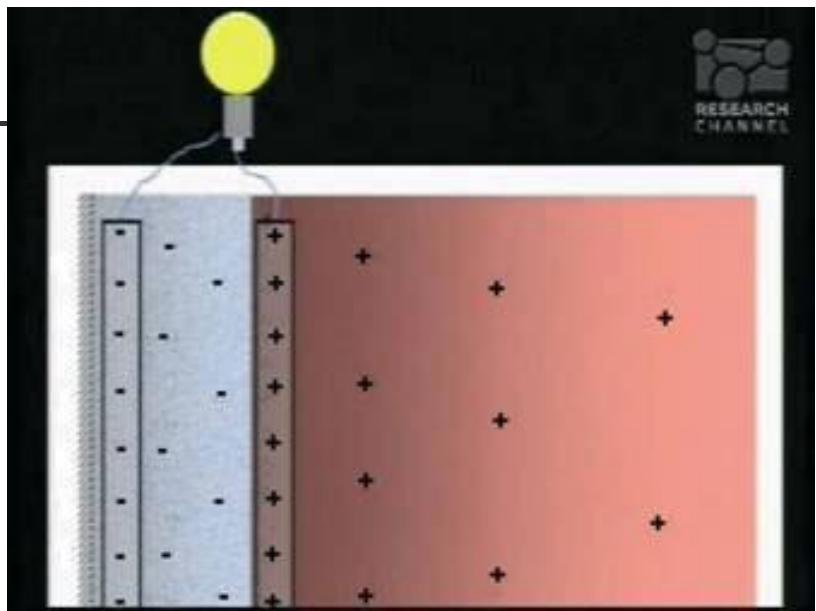
H⁺

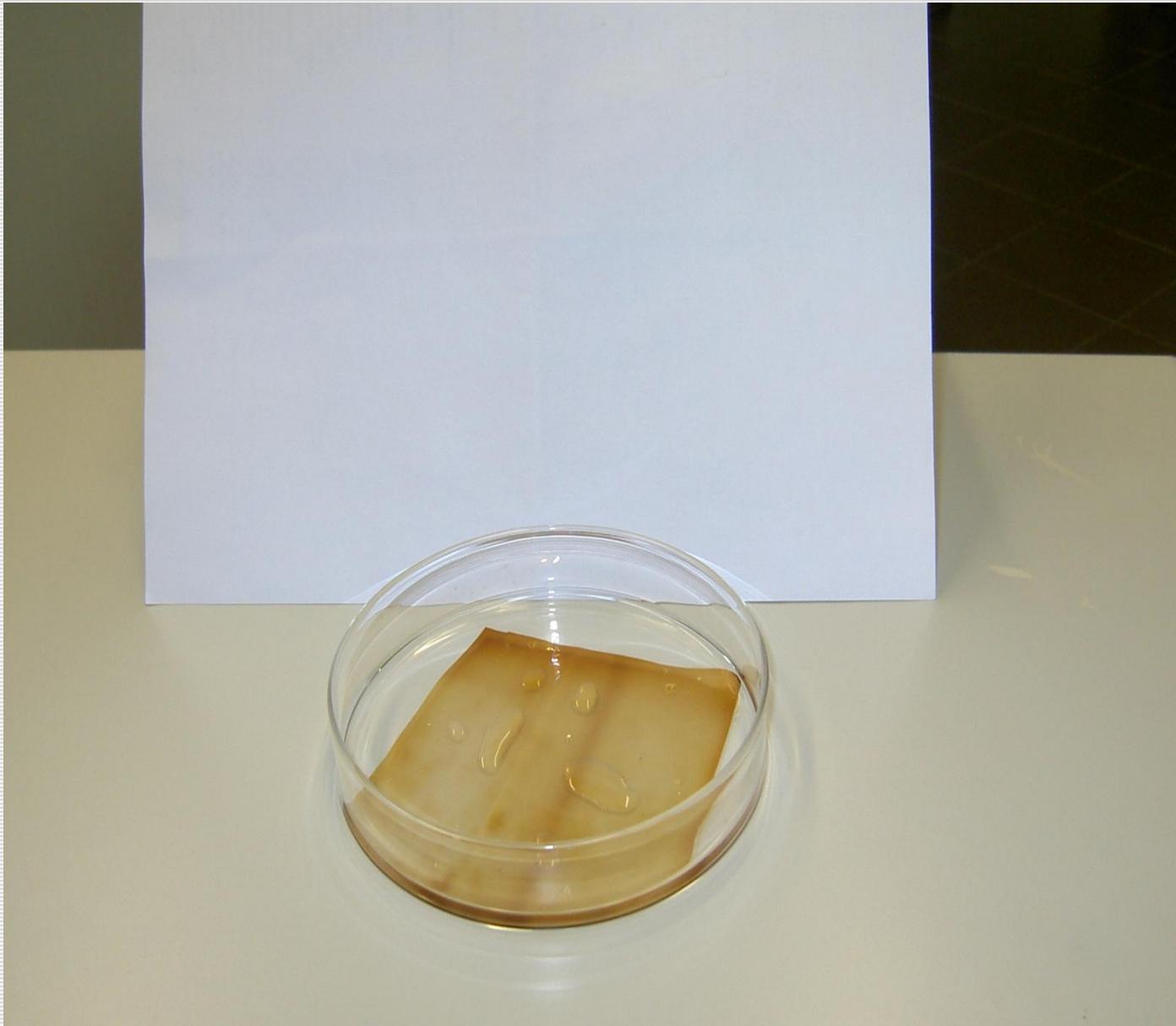
Nafion

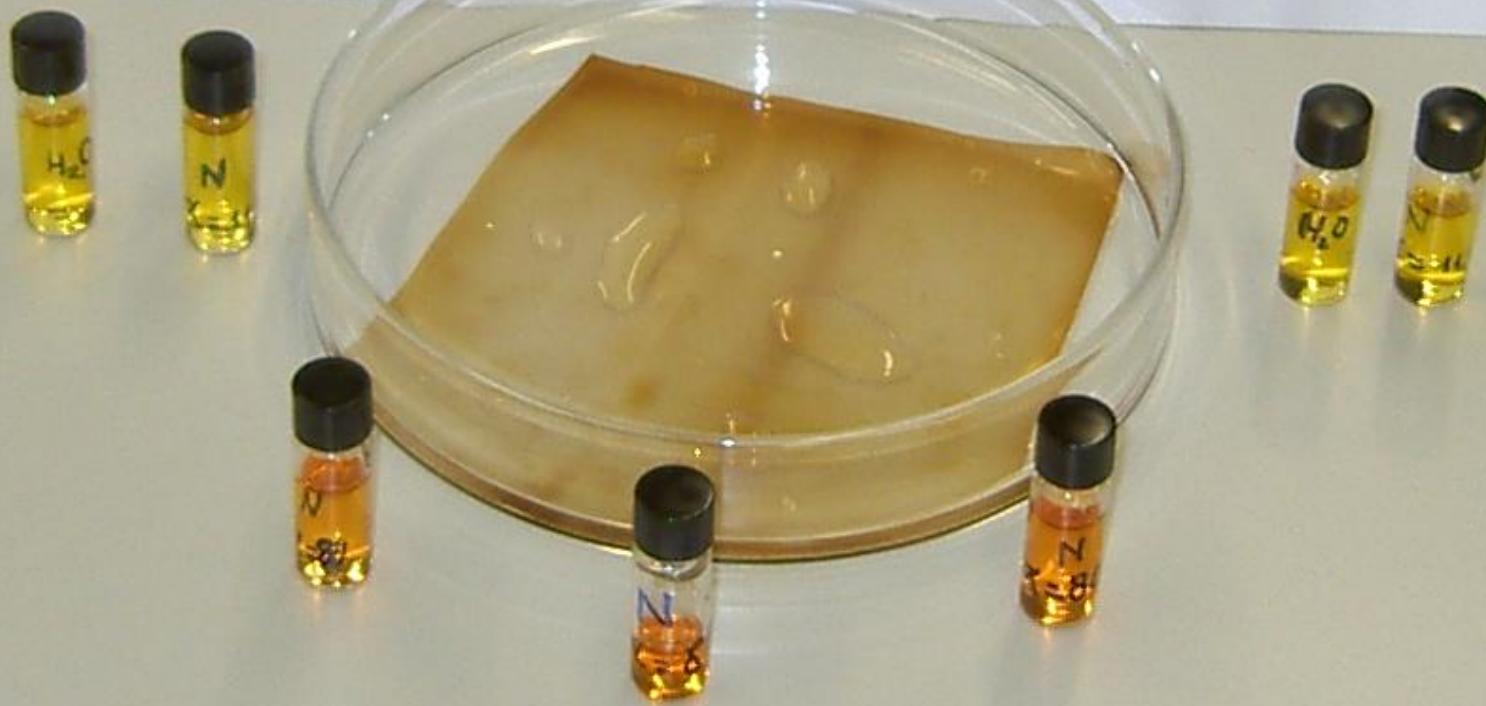
EZ dye excluded

1 μm



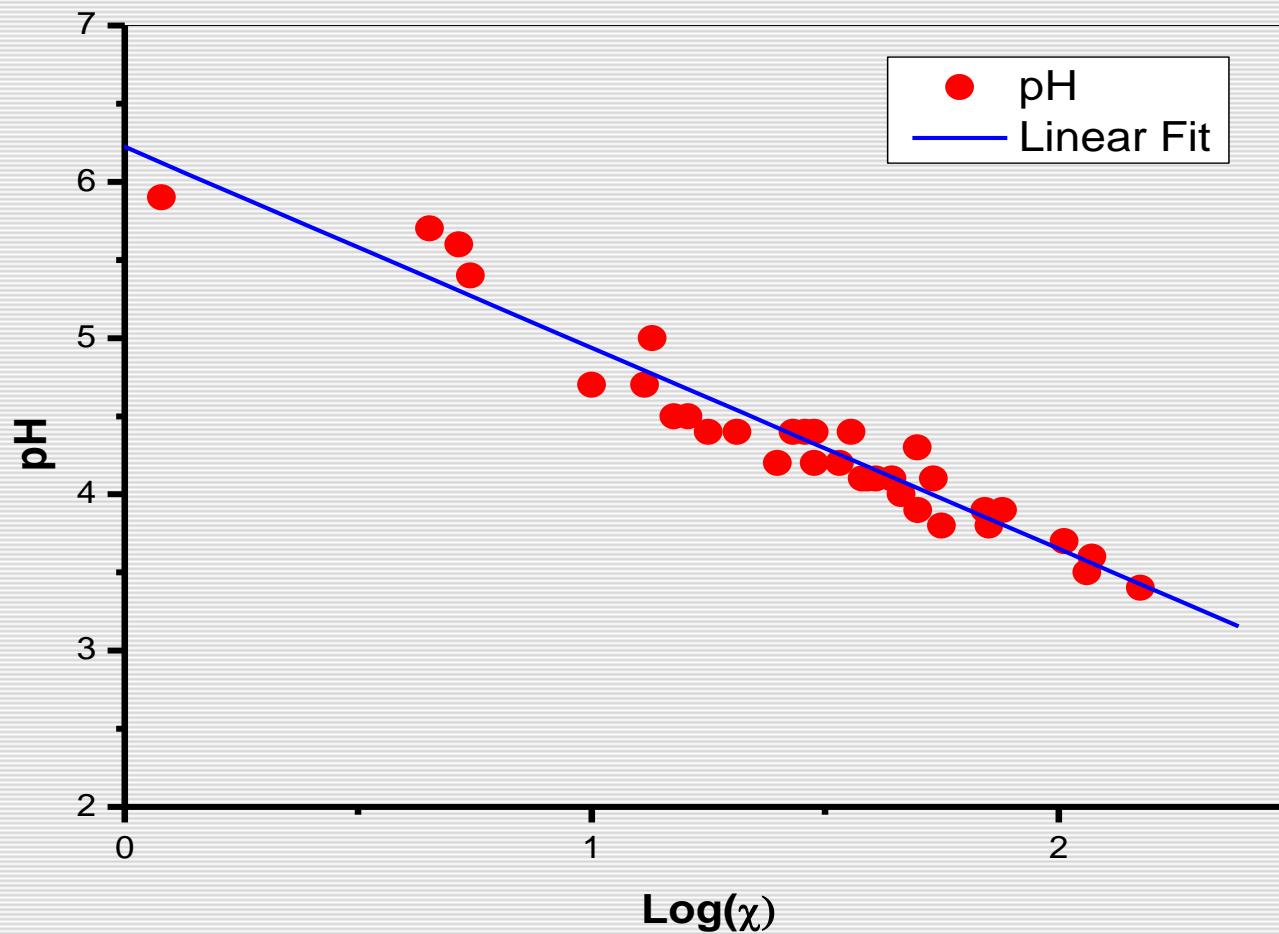












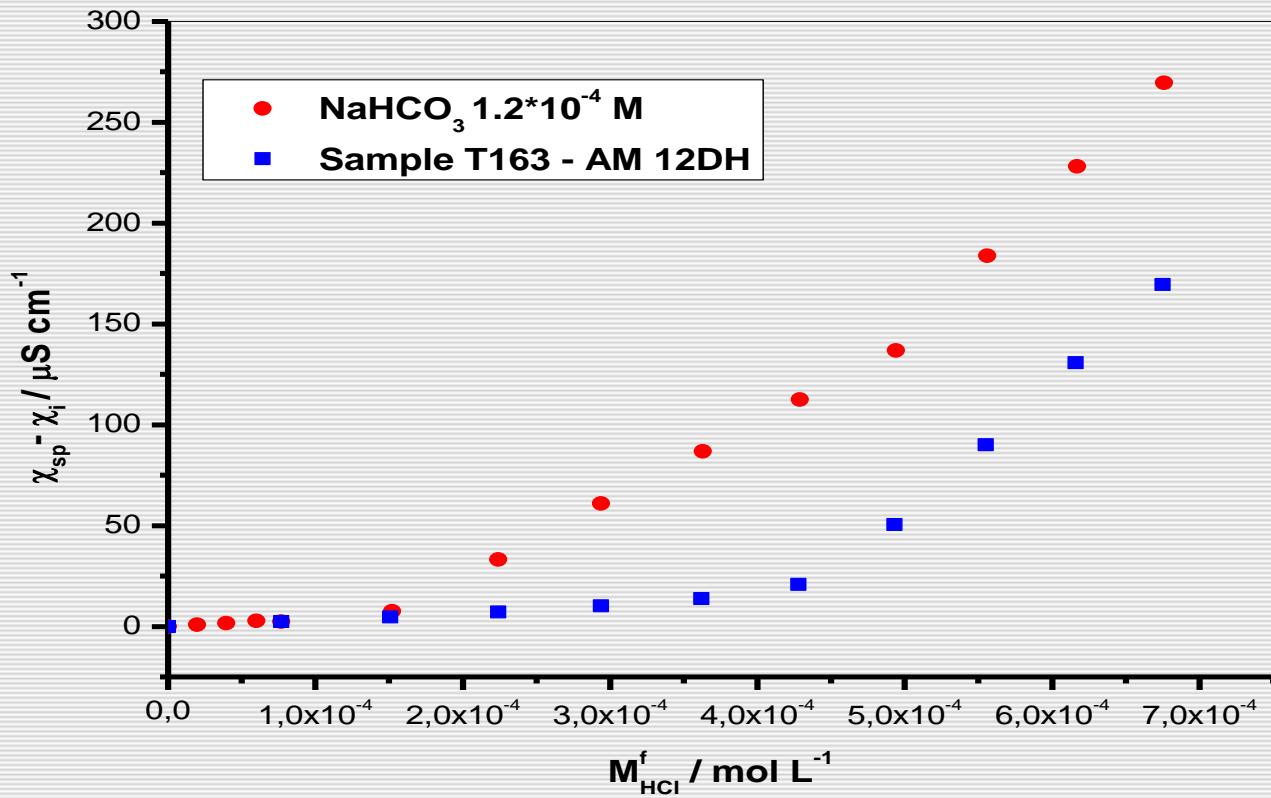


Titrations:

- *Conductometric*
- *pHmetric*
- *Calorimetric*



Conductometric Titration with HCl as titrant



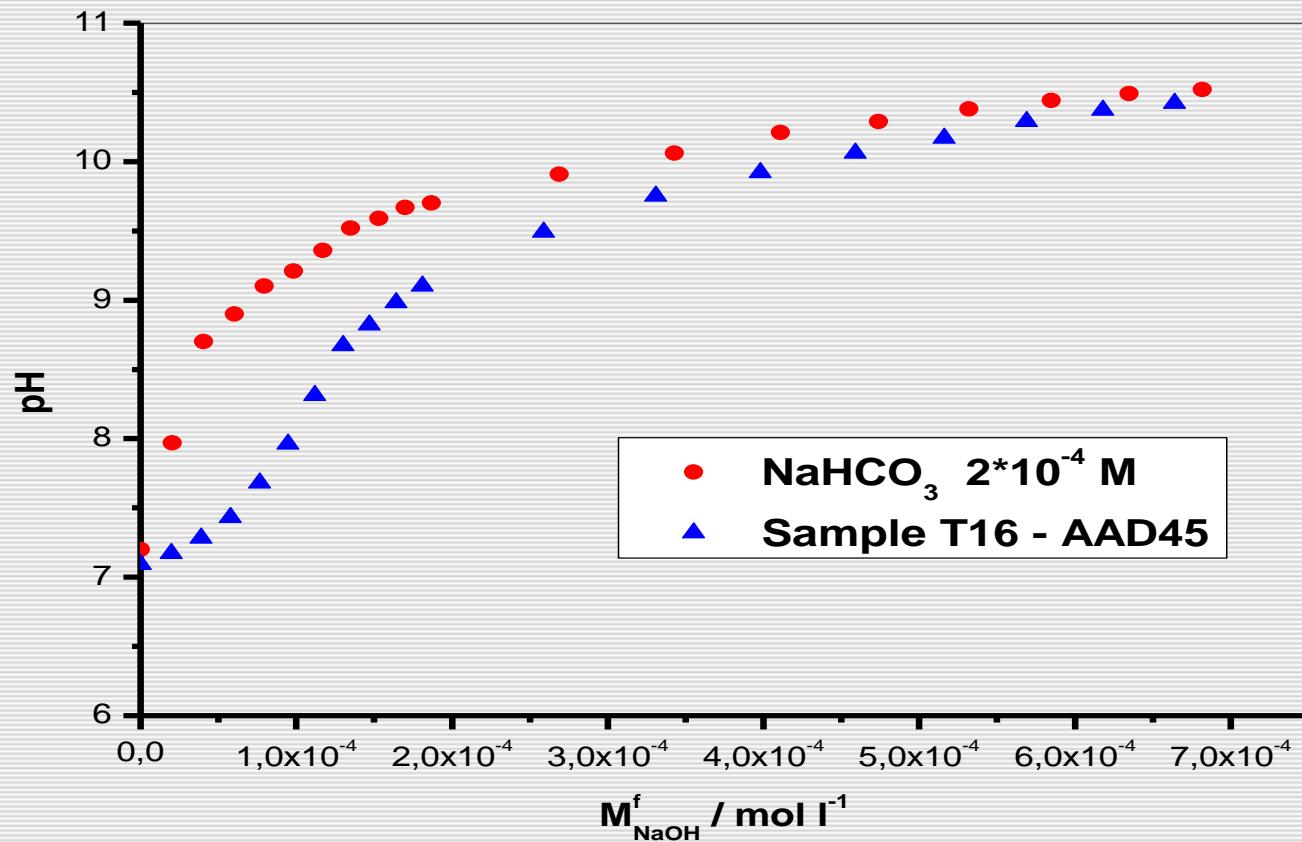
C.M.Cacace, L.Elia, V.Elia, E.Napoli, M.Niccoli
Journal Molecular Liquids, 2009, 146, 122-126

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pHmetric Titration with NaOH as titrant



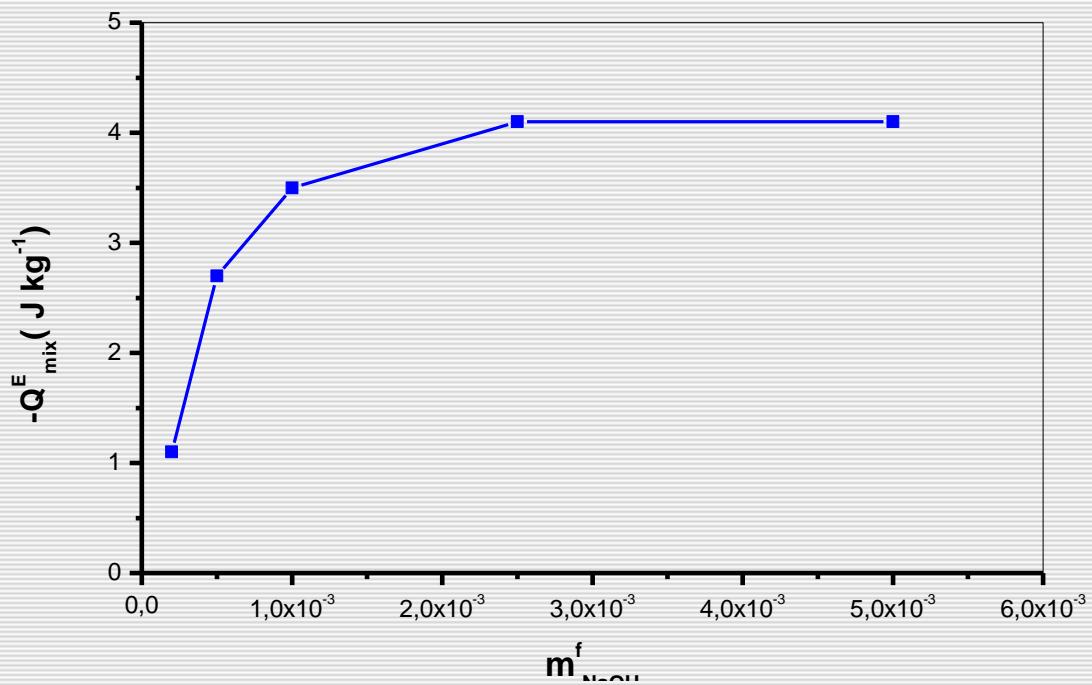
V.Elia, E. Napoli, M. Niccoli

Journal of Molecular Liquids, 2009 , Vol 149, 45-50

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Dipartimento

Calorimetric Titration with NaOH as titrant



V.Elia,M.Marchese,M.Montanino,E.Napoli,M.Niccoli,L.Nonatelli, and A.Ramaglia
Journal of Solution chemistry, 2005 Vol.34 (8)

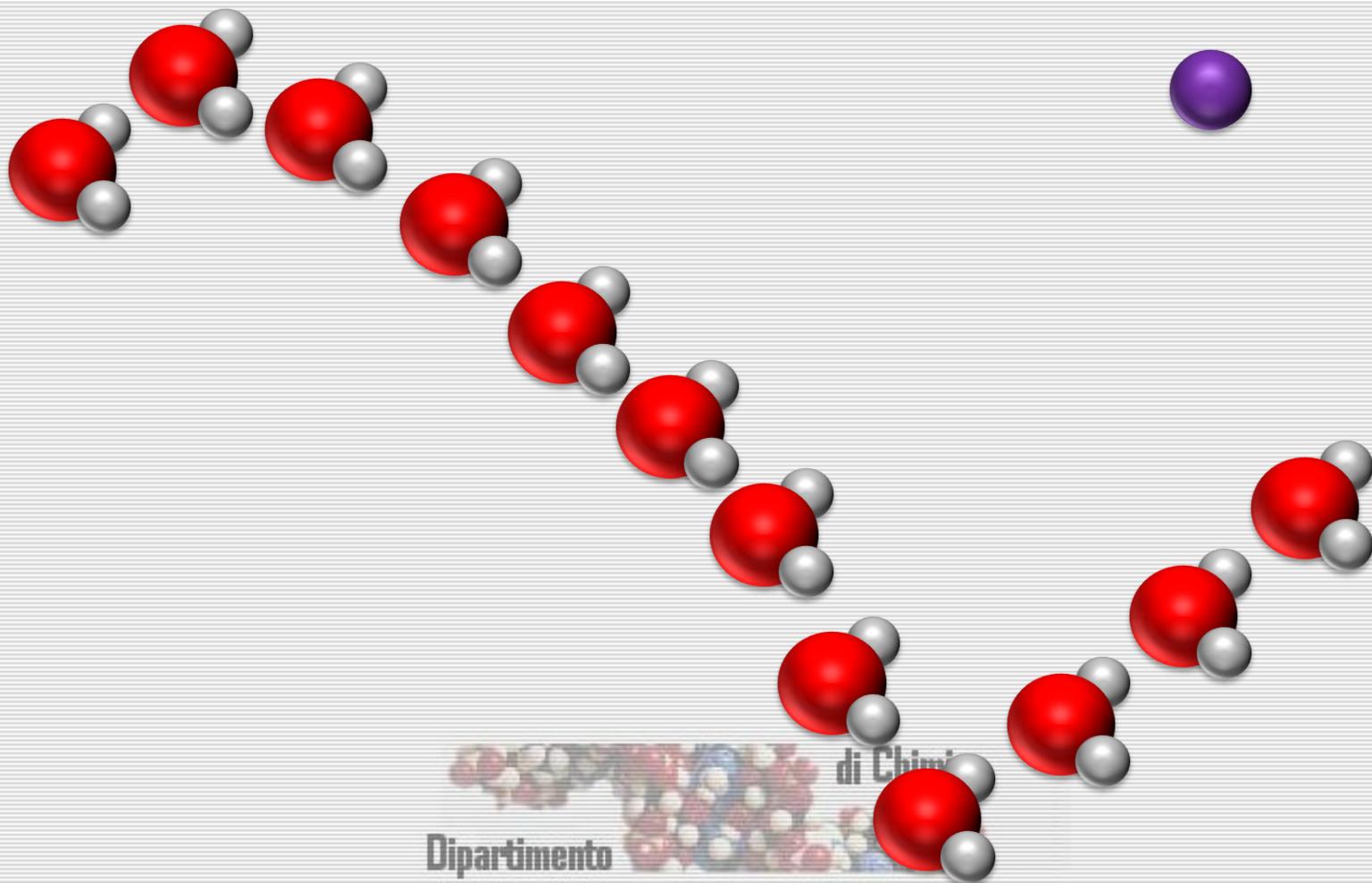
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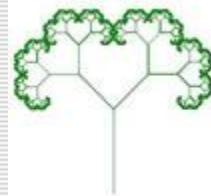
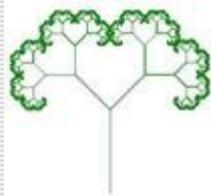
Dipartimento



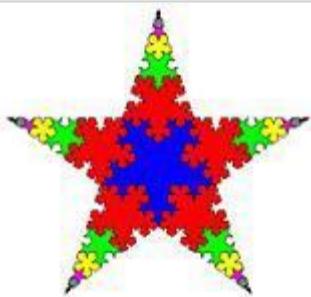
Thermodynamic parameters for the association between DS and OH ions at 298 K

System	K' Kg mol^{-1}	$-\Delta H^\circ$ kJ mol^{-1}	$-\Delta G^\circ$ kJ mol^{-1}	$-\Delta G^\circ'$ kJ mol^{-1}	T ΔS° kJ mol^{-1}	M _{DS} *10 ⁴ mol L^{-1}	M _{Na+} *10 ⁴ mol L^{-1}
Tmix ₁	2612 \pm 1170	188 \pm 32	19 \pm 1	19 \pm 1	-168 \pm 33	1,2	1,6
Tmix ₂	1876 \pm 615	80 \pm 18	18.8 \pm 0.8	18.8 \pm 0.8	-61 \pm 19	3,0	1,7
Mix ₃	1986 \pm 519	184 \pm 25	18.8 \pm 0.7	18.8 \pm 0.7	-165 \pm 26	2,0	3,1
Mix ₁₂₂	8977 \pm 877	70 \pm 2	22.6 \pm 0.2	22.6 \pm 0.2	-47 \pm 2	2,8	7,1





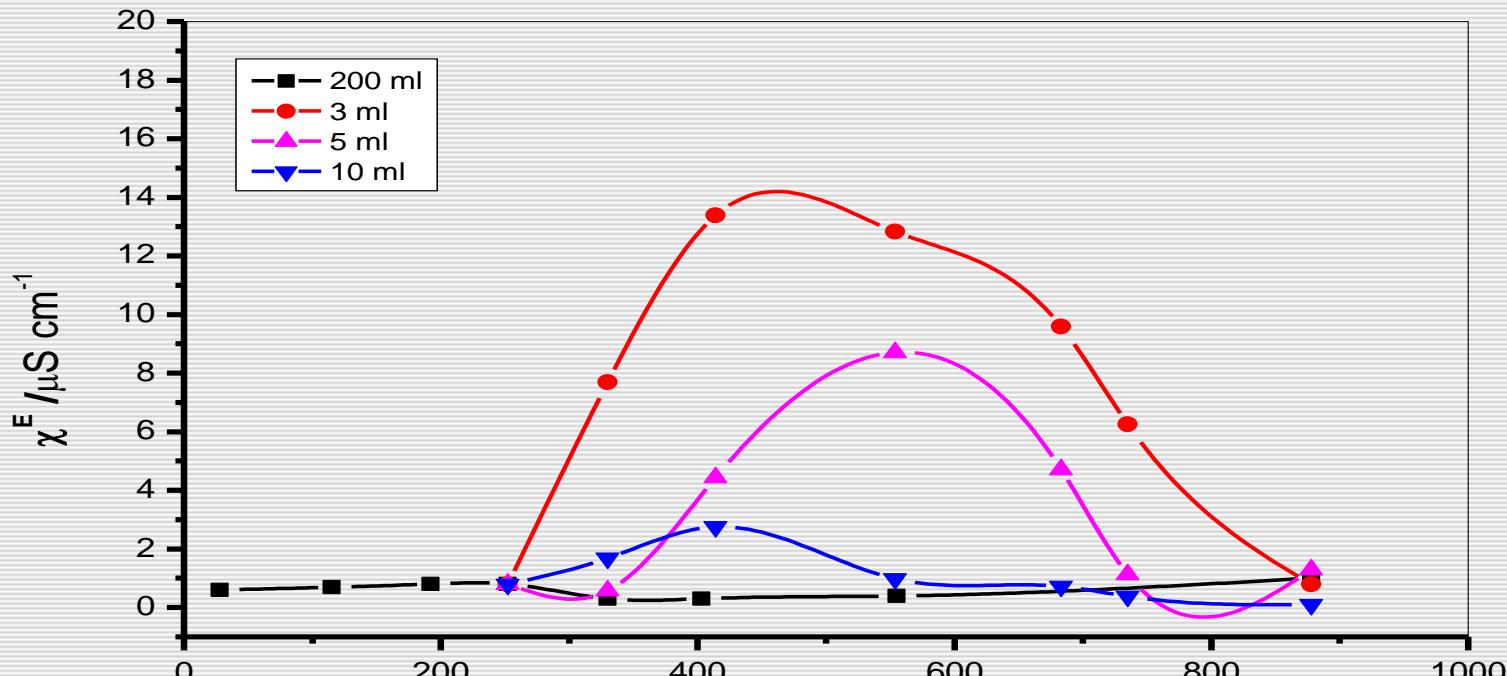
DISSIPATIVE STRUCTURES





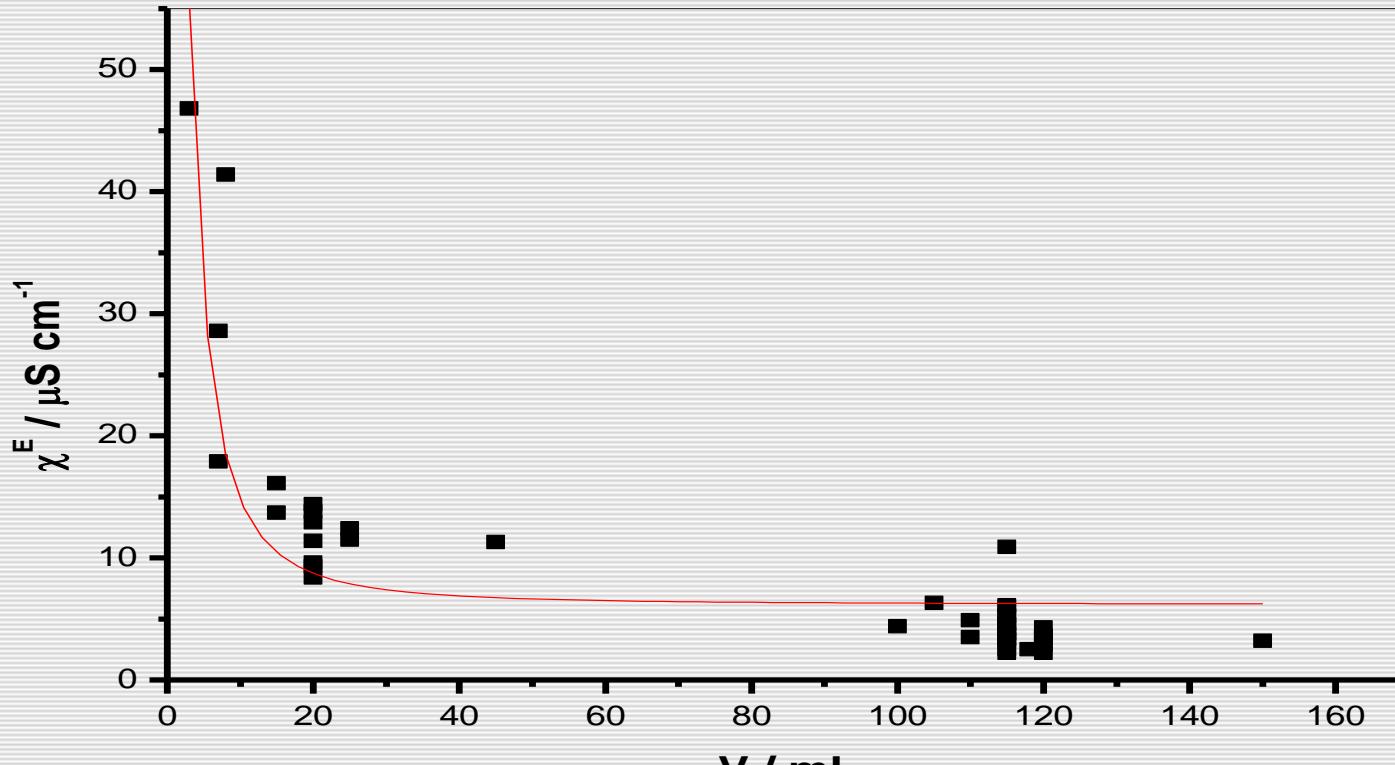
Dipartimento di Chimica

Combined Volume – Time Effect



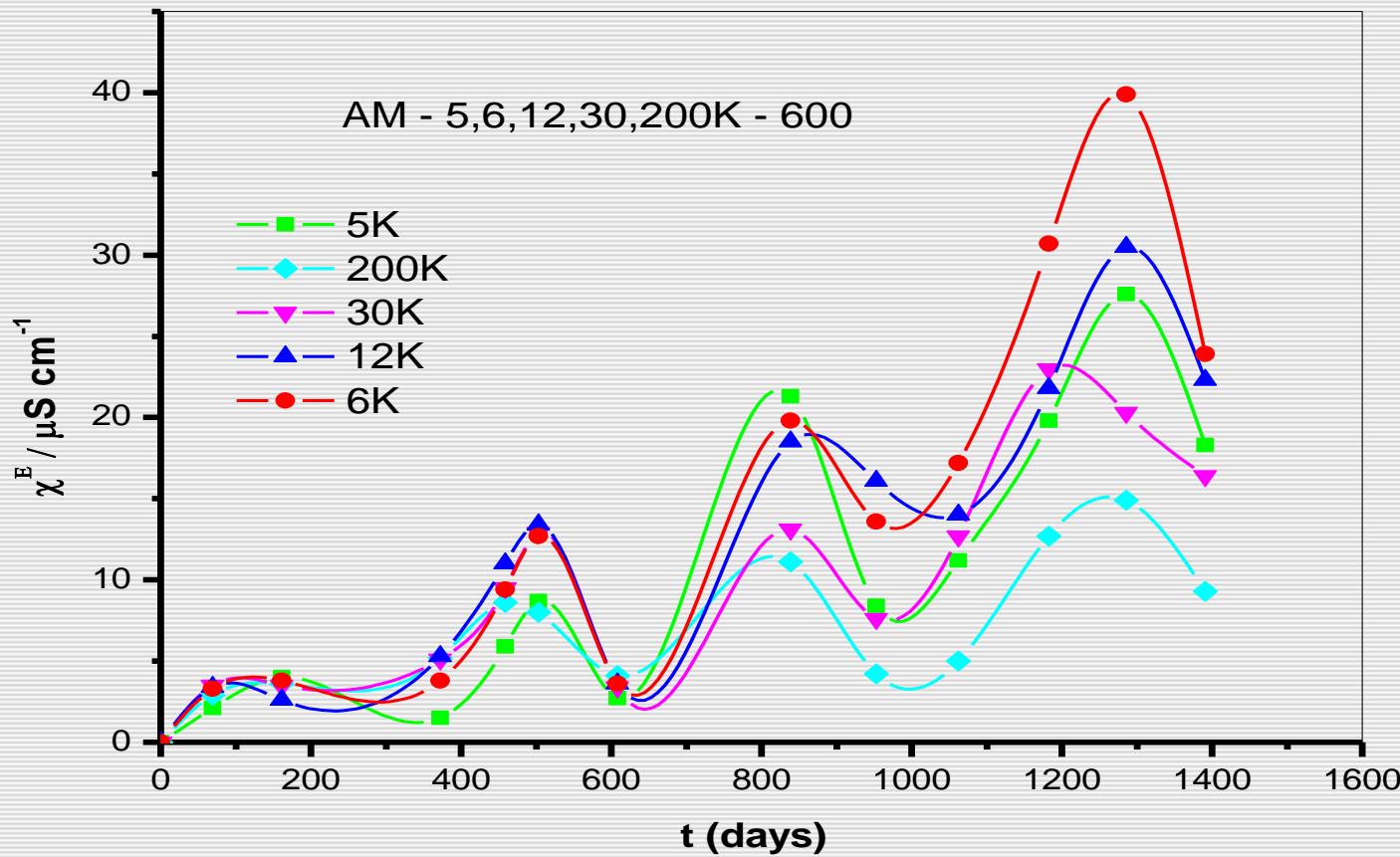
P.Belon, V. Elia, L. Elia, M. Montanino, E. Napoli, M. Niccoli
Journal of Thermal Analysis and Calorimetry, (2008), Vol.93 (2), 459-469

Excess Specific Electric Conductivity vs. Volume



V. Elia, L. Elia, E. Napoli, M. Niccoli
“International Journal of Ecodynamics, Vol.1 No.4 (2007)

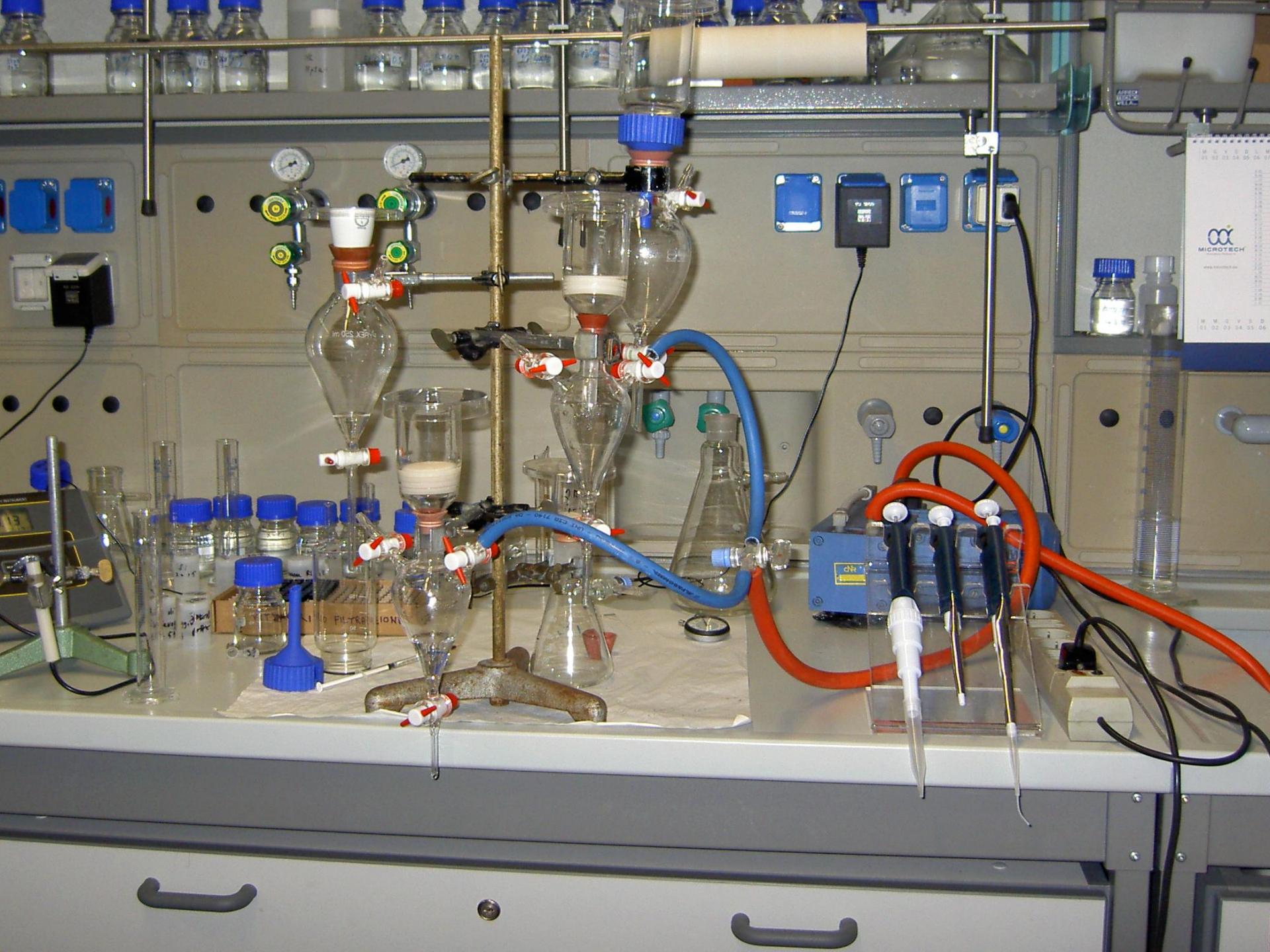
Excess Conductivity vs time of AM 5,6,12,30,200CK

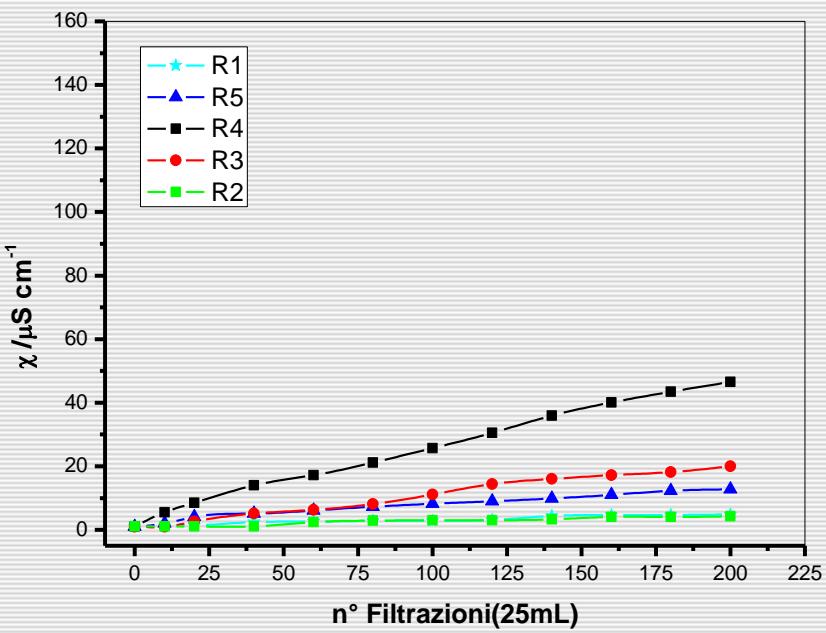
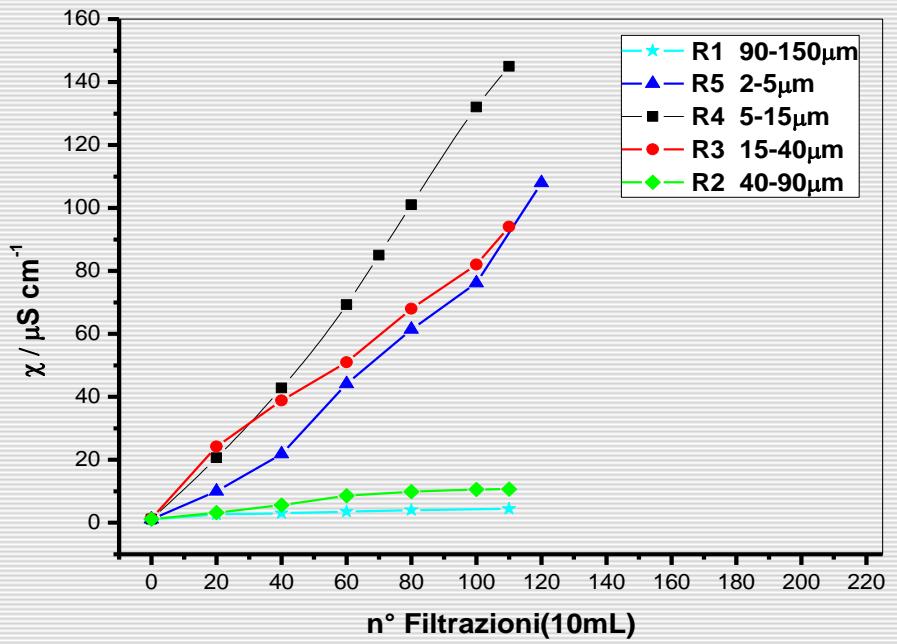


V.Elia,L.Elia,N. Marchettini, E.Napoli, M.Niccoli, E. Tiezzi
Journal of Thermal Analysis and Calorimetry, 2008, Vol. 93 (3), 1003

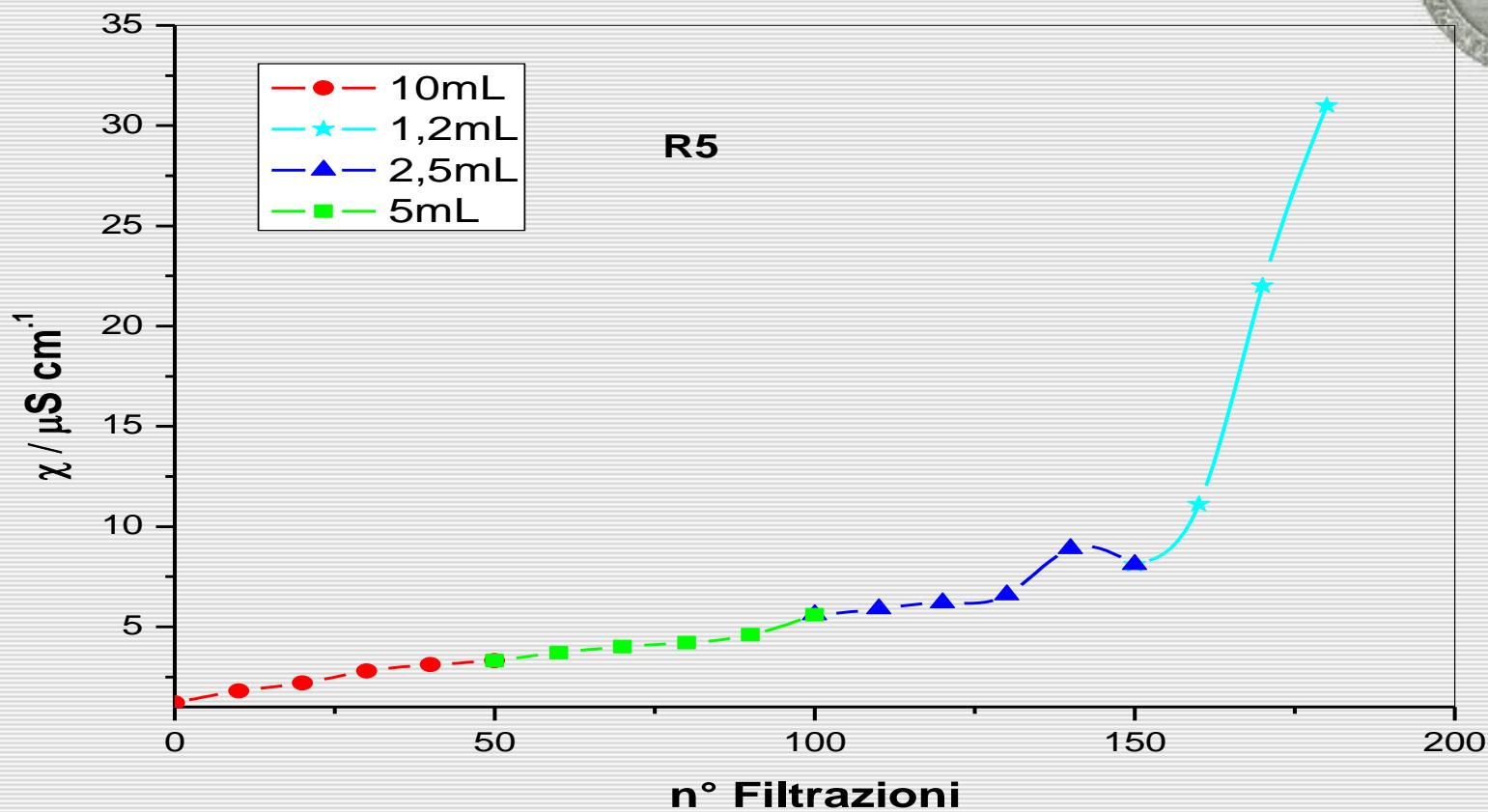


*The Effect of Filtration
Procedures on the
Sovramolecular Structure
of Water*





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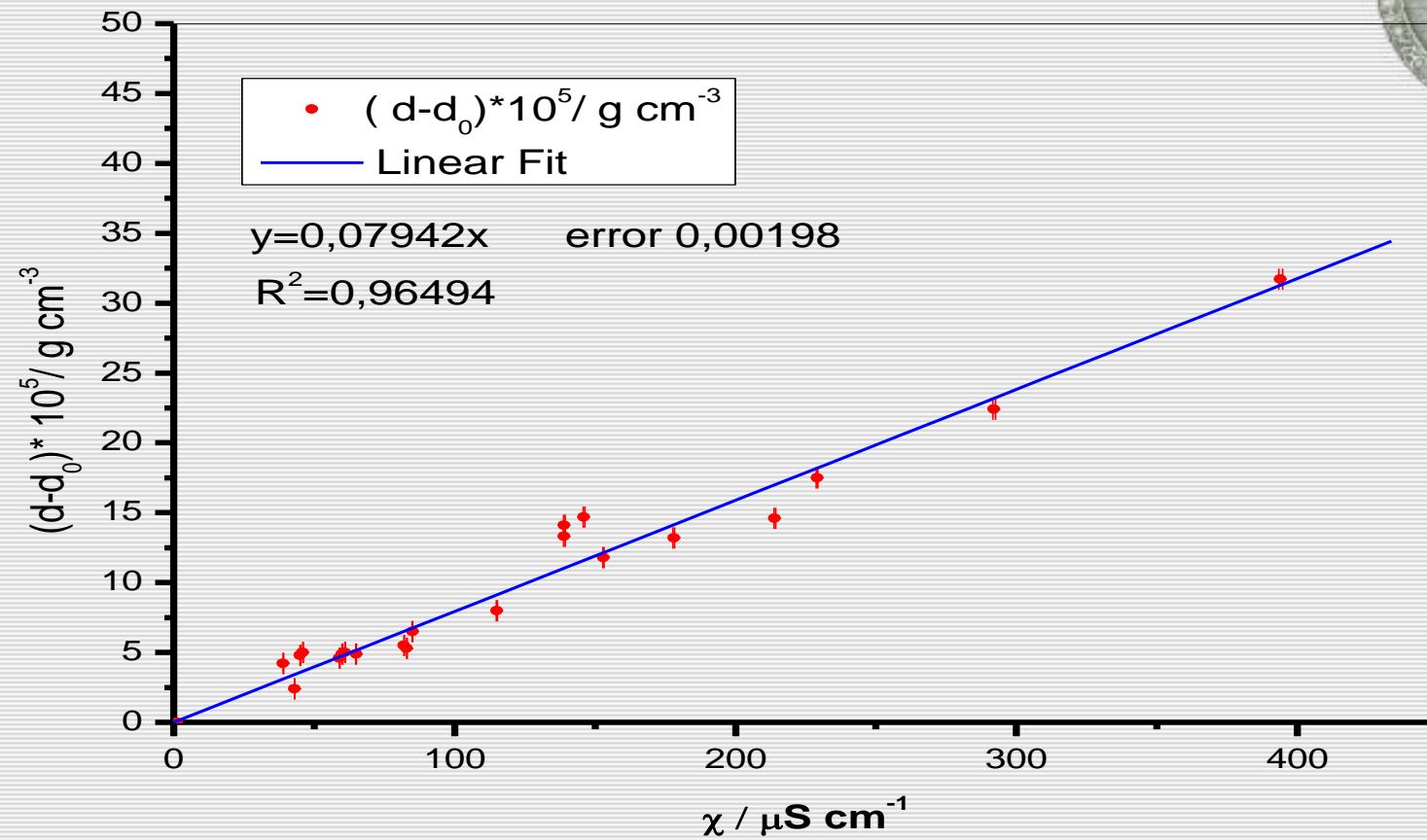


Fig.8 - Density vs Specific Electrical Conductivity (See Table 8)



Quando la scienza si arrocca su un certo paradigma, escludendo come pazzo o eretico chi lo contesta, si comporta in modo dogmatico.

Come la mettiamo col fatto che l'innovazione avviene proprio quando qualcuno riesce a mettere in questione il paradigma dominante?

Umberto Eco



*“If we knew what it was we
were doing, it would not be
called research, would it?”*

A. Einstein



Thank you for your attention